



A Publication  
of Reliable Methods  
for the Preparation  
of Organic Compounds

## Working with Hazardous Chemicals

The procedures in *Organic Syntheses* are intended for use only by persons with proper training in experimental organic chemistry. All hazardous materials should be handled using the standard procedures for work with chemicals described in references such as "Prudent Practices in the Laboratory" (The National Academies Press, Washington, D.C., 2011; the full text can be accessed free of charge at [http://www.nap.edu/catalog.php?record\\_id=12654](http://www.nap.edu/catalog.php?record_id=12654)). All chemical waste should be disposed of in accordance with local regulations. For general guidelines for the management of chemical waste, see Chapter 8 of Prudent Practices.

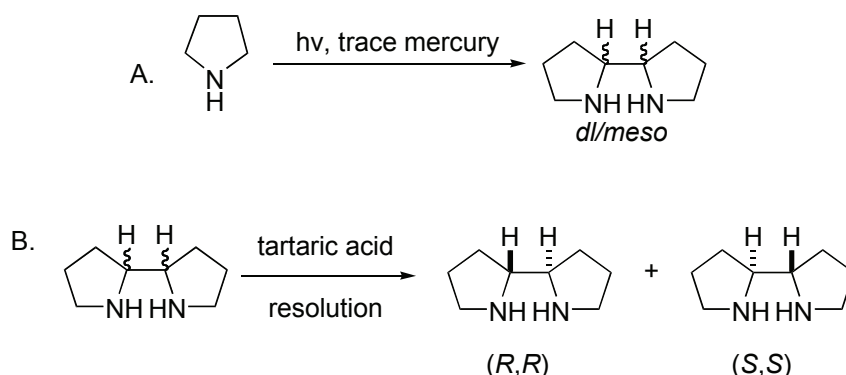
In some articles in *Organic Syntheses*, chemical-specific hazards are highlighted in red "Caution Notes" within a procedure. It is important to recognize that the absence of a caution note does not imply that no significant hazards are associated with the chemicals involved in that procedure. Prior to performing a reaction, a thorough risk assessment should be carried out that includes a review of the potential hazards associated with each chemical and experimental operation on the scale that is planned for the procedure. Guidelines for carrying out a risk assessment and for analyzing the hazards associated with chemicals can be found in Chapter 4 of Prudent Practices.

The procedures described in *Organic Syntheses* are provided as published and are conducted at one's own risk. *Organic Syntheses, Inc.*, its Editors, and its Board of Directors do not warrant or guarantee the safety of individuals using these procedures and hereby disclaim any liability for any injuries or damages claimed to have resulted from or related in any way to the procedures herein.

*September 2014: The paragraphs above replace the section "Handling and Disposal of Hazardous Chemicals" in the originally published version of this article. The statements above do not supersede any specific hazard caution notes and safety instructions included in the procedure.*

Copyright © 2006 Organic Syntheses, Inc. All Rights Reserved

**(*R,R*)-2,2'-BISPYRROLIDINE and (*S,S*)-2,2'-BISPYRROLIDINE:  
USEFUL LIGANDS FOR ASYMMETRIC SYNTHESIS**

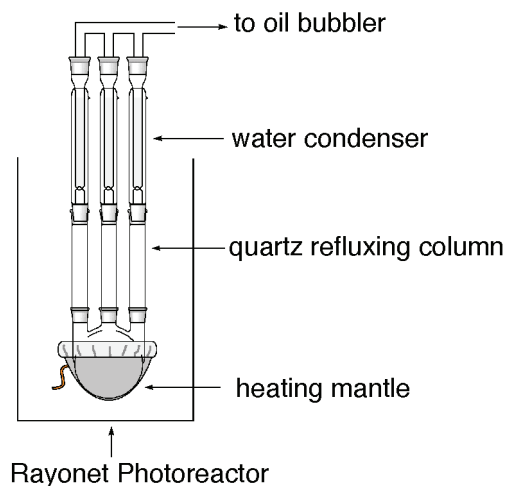


Submitted by Scott E. Denmark, Jiping Fu and Michael J. Lawler.<sup>1</sup>  
Checked by Sandra Lee, Elliott Huntsman and Edward J.J. Grabowski.

*Caution! This procedure should be carried out in a well-ventilated hood because of the stench of pyrrolidine and 2,2'-bispyrrolidine. The hood doors should be covered with an opaque sheet to shield the UV light.*

### 1. Procedure

A. (*R\*,R\**) and (*R\*,S\**)-2,2'-Bispyrrolidine.<sup>2,3</sup> A three-necked, 500-mL flask equipped with three quartz refluxing columns and three water condensers (Figure 1) is charged with pyrrolidine (160 mL) and a drop of mercury (Notes 1 and 2). The water condensers are equipped with gas inlets connected to a single nitrogen source leading to an oil bubbler. The flask is placed in a Rayonet Photoreactor fitted with 14 x 8 Watt low pressure Hg lamps (254 nm). The reaction mixture is heated to reflux with a heating mantle. The lamps are turned on, after which the mixture is heated at reflux for 7 days. After the lamps are turned off, the system is allowed to cool to room temperature (Note 2). The liquid is then carefully decanted to a distillation flask and the mercury is recovered. Unreacted pyrrolidine and side products are removed by distillation at atmospheric pressure (Note 3). The residue is distilled to provide 67.6 g (50%) of a mixture of (*R\*,R\**) and (*R\*,S\**)-2,2'-bispyrrolidine as a clear, light yellow liquid (bp 79–81 °C at 3.0 mmHg) (Note 4). The product is of sufficient purity to be used in the resolution step (Note 5)



**Figure 1.** Apparatus for photodimerization of pyrrolidine.

### B. Resolution of 2,2'-Bispyrrolidine<sup>3,4</sup>

1. *Preparation of (R,R)-2,2'-bispyrrolidine•(L)-tartrate.* To a solution of a 1/1 mixture of *d,l*- and *meso*-2,2'-bispyrrolidine (67.0 g, 479 mmol) in H<sub>2</sub>O (240 mL) is added (L)-(+)-tartaric acid (36.0 g, 240 mmol, 0.5 equiv) and acetic acid (27.4 mL, 479 mmol, 1.0 equiv) (Note 6). The mixture is heated to 90 °C and the homogenous solution is allowed to cool to room temperature slowly before it is placed in an ice bath. After the solid precipitates (Note 7), the mixture is kept in an ice bath for another 2 h. The precipitate is then filtered and is washed with ice-cold water (20 mL). The mother liquor is saved for the recovery of the (*S,S*)-2,2'-bispyrrolidine. The collected solid is dried under high vacuum (0.5 mmHg) at 80 °C for 2 h to give 31.6 g of a light yellow powder. This solid is dissolved in hot water (90 mL) and the solution is allowed to cool slowly to room temperature before it is placed in an ice bath for 1 h. The precipitate is filtered and the solid is washed with ice-cold water (10 mL). The solid is dried under high vacuum (0.5 mmHg) at 80 °C for 2 h to give 23.0 g of white crystals. This recrystallization process is repeated using water (50 mL) to give 21.4 g (62% based on isomer content) of (*R,R*)-2,2'-bispyrrolidine•(L)-tartrate as white prismatic crystals (Note 8).

2. *Preparation of (R,R)-2,2'-Bispyrrolidine.* To a mixture of the tartrate salt (9.1 g) in water (15 mL) at 0 °C are added KOH (20 g) pellets (Note 9). The mixture is then stirred at 0 °C for 10 min before diethyl ether

(80 mL) is added, whereupon the solution is stirred at 0 °C for another 30 min. The aqueous layer is then separated and extracted with diethyl ether (6 x 50 mL) (Note 10). The diethyl ether extracts are combined, dried (K<sub>2</sub>CO<sub>3</sub>) and then are concentrated under vacuum (Note 11). The residue is transferred to a dry 25-mL, round-bottomed flask. To this is added a small piece of sodium (Note 12) and the mixture is stirred at room temperature under nitrogen for 30 min. The residue is distilled under vacuum to give 3.61 g (83%) of (*R,R*)-2,2'-bispyrrolidine as a clear colorless oil (bp 97–98 °C at 8.0 mmHg) (Note 13). The enantiomeric purity of product is determined by CSP-SFC and CSP-HPLC analysis of the corresponding dibenzoyl amide derivative (Note 14).

3. *Preparation of (S,S)-2,2'-bispyrrolidine•(D)-tartrate.* The mother liquor from initial resolution is cooled to 0 °C and KOH pellets (80 g) are added slowly. The mixture is stirred vigorously at 0 °C for 10 min (Note 15). To this solution is added diethyl ether (500 mL) and the mixture is stirred at room temperature for 20 min. The aqueous layer is separated and then is extracted with diethyl ether (4 x 500 mL). The diethyl ether extracts are combined, dried (K<sub>2</sub>CO<sub>3</sub>), and then are concentrated under vacuum to give 48.4 g of a yellow oil. The oil is dissolved in H<sub>2</sub>O (150 mL), then (D)-(-)-tartaric acid (34.5 g, 230 mmol) and acetic acid (27.0 mL, 473 mmol) are added (Note 16). The mixture is heated to 90 °C and the homogenous solution is allowed to cool to room temperature slowly before it is cooled in an ice bath. After the solid precipitates, the mixture is kept in an ice bath for another 2 h. The precipitate is filtered and the solid is washed with ice-cold water (10 mL), and then is dried under high vacuum (0.5 mmHg) at 80 °C for 2 h to give 23.5 g of a light-yellow powder. The solid is dissolved in hot water (60 mL) and the solution is allowed to cool slowly to room temperature before it is placed in an ice bath for another 2 h. The precipitate is filtered, washed with 10 mL of ice-cold water, and dried under high vacuum (0.5 mmHg) at 80 °C for 2 h to give 20.16 g of white prismatic crystals. This recrystallization process is repeated using water (55 mL) to give 18.4 g (55% based on isomer content) of (*S,S*)-2,2'-bispyrrolidine•(D)-(-)-tartrate as white prismatic crystals (Note 17).

4. *Preparation of (S,S)-2,2'-Bispyrrolidine.* To a solution of the tartrate salt (9.3 g) in water (15 mL) at 0 °C are added KOH pellets (20 g). The mixture is stirred at 0 °C for 10 min before diethyl ether (80 mL) is added, whereupon it is stirred at 0 °C for another 30 min. The aqueous layer is separated and then is extracted with diethyl ether (6 x 50 mL). The

diethyl ether extracts are combined, dried ( $K_2CO_3$ ) and then are concentrated *in vacuo* (Note 11). The residue is then transferred to a dry 25-mL, round-bottomed flask. To this is added a piece of sodium (Note 12) and the mixture is stirred at room temperature under nitrogen for 30 min. The residue is then distilled under vacuum to give 3.51 g (80%) of (*S,S*)-2,2'-bispyrrolidine as a clear, colorless oil (bp 97–98 °C at 8.0 mmHg) (Note 18). The enantiomeric purity of product is determined by CSP-SFC and CSP-HPLC analysis of corresponding dibenzoyl amide derivative (Note 19).

## 2. Notes

1. The three quartz-refluxing columns are 34 cm long and 4.0 cm in diameter. All joints are well sealed with high vacuum grease to avoid leakage. Pyrrolidine (99 %) was purchased from Aldrich Chemical Company, Inc., and was used without further purification.

Variables that are difficult to control relative to the photolysis include the age and actual power of the Hg lamps, the actual length of the quartz tubes receiving the UV light and the rate of reflux. All affect the overall rate of photolysis. The checkers found it convenient to follow the progress of the photolysis by periodically sampling the reaction and analyzing it by  $^{13}C$ -NMR spectroscopy in  $CDCl_3$ . The peak heights for the methylene groups for pyrrolidine (47.1 ppm) and the bis-pyrrolidines (47.0 and 46.6) were used as measures of the relative ratios of these species. When the pyrrolidine is almost consumed in the photolysis, the reflux ceases. The submitters were able to recover ~50% of the bis-pyrrolidines after distillation following a seven-day photolysis (See Note 5). The checkers achieved the following results: 39% in seven days; 60% in nine days; 72% in 11.4 days and 60% in five days at half-scale.

2. The crude reaction mixture can be analyzed by  $^1H$  NMR. The ratio of starting material to products can be estimated from the NMR spectrum.

3. The distillate contains 46.6 g of colorless liquid.

4. The distillation should be done carefully to avoid solidification of the diamine in the condenser.

5. The crude material (67.6 g, approximately 50% based on the pyrrolidine charged) contains a ca. 1/1 mixture of *d,l* and *meso* isomers:  $^1H$  NMR (500 MHz,  $CDCl_3$ ) d: 1.32–1.46 (m, 2 H), 1.61–1.92 (m, 8 H), 2.82–2.98 (m, 6 H). The checkers obtained 52.6 g from the seven-day photolysis; 81.0 g from the nine day photolysis; 97.0 g from the 11.4 day

photolysis and 41.5 g from the five day photolysis at half-scale.

6. L-(+)-Tartaric acid (99% GLC) was purchased from Aldrich Chemical Company, Inc., and was used without further purification. Acetic acid (glacial) was purchased from Fisher Scientific Company and was used without further purification. In completing the checking of this procedure, the subsequent reactions were scaled to reflect the quantity of bispyrrolidines obtained in the distillations. The reactions checked at the yields indicated.

7. The initial formation of crystals may take up to 16 h. The process can be facilitated by stirring the mixture with glass rod or by addition of small amount of seed crystals.

8. The analytical data for (*R,R*)-2,2'-bispyrrolidine•(L)-tartrate are as follows:<sup>4</sup> mp 212–216 °C; <sup>1</sup>H NMR (500 MHz, D<sub>2</sub>O/DSS) δ: 1.88–1.98 (m, 2 H), 2.08–2.28 (m, 4 H), 2.40–2.48 (m, 2 H), 3.51–3.55 (m, 4 H), 3.92–4.00 (m, 2 H), 4.43 (s, 2 H), 4.86 (br, 6 H); <sup>13</sup>C NMR (126 MHz, D<sub>2</sub>O) δ: 25.5, 31.0, 49.1, 63.2, 76.6, 181.4; IR (KBr) cm<sup>-1</sup>: 3220, 2717, 2516, 1693, 1612, 1583, 1450, 1386, 1321, 1124, 1072, 709; [α]<sub>D</sub><sup>24</sup> +17.9 (*c* = 1.00, H<sub>2</sub>O); Anal. Calcd for C<sub>12</sub>H<sub>22</sub>N<sub>2</sub>O<sub>6</sub>: C, 49.65; H, 7.64, N, 9.65. Found: C, 49.80; H, 7.63; N, 9.65. The checkers noted slight chemical shift variations in the NMR spectra of this material, and attribute these to slight differences in concentration and apparent pH in the different samples.

9. Potassium hydroxide (87.7%) was purchased from Fisher Scientific Company and was used without further purification.

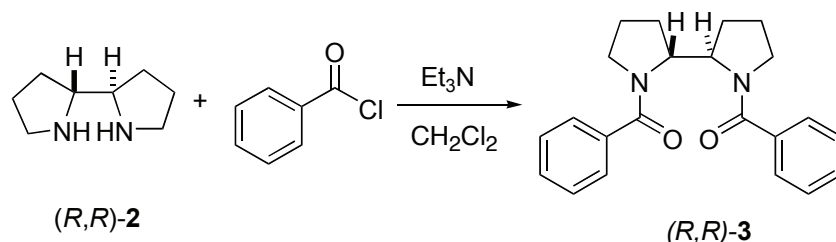
10. Diethyl ether was purchased from Mallinckrodt Inc. and was used without purification.

11. The water bath is kept at 0 °C to avoid loss of (*R,R*)-2,2'-bispyrrolidine.

12. The piece of sodium is about 0.5 cm<sup>3</sup> and it is washed with hexane before use. After distillation, the sodium was destroyed by the careful addition of isopropyl alcohol.

13. The analytical data for (*R,R*)-2,2'-bispyrrolidine are as follows:<sup>3,4</sup> <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ: 1.31–1.38 (m, 2 H), 1.65–1.84 (m, 6 H), 2.06 (br, 2 H), 2.82–2.97 (m, 6 H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ: 25.4, 29.0, 46.4, 63.8; IR (KBr) cm<sup>-1</sup>: 3270, 2956, 2867, 1282, 1118, 1076; MS (FAB) (*m/z*): 141; HRMS (*m/z*) C<sub>8</sub>H<sub>17</sub>N<sub>2</sub> (M+H): Calc.: 141.1386; Found: 141.1375; [α]<sub>D</sub><sup>24</sup> -14.91 (*c* = 1.03, MeOH); Anal. Calcd for C<sub>8</sub>H<sub>16</sub>N<sub>2</sub>: C, 68.52; H, 11.50, N, 19.98. Found: C, 68.38; H, 11.64; N, 19.92. The free diamine is extremely hygroscopic, oxygen sensitive and absorbs CO<sub>2</sub> rapidly in air.

14. Procedure for derivatization is as follows (eq 1)<sup>3,4</sup>: To a solution of (*R,R*)-**2** (140 mg, 1.0 mmol) in 1.0 mL of methylene chloride (purchased from Fisher Scientific Company and distilled from P<sub>2</sub>O<sub>5</sub>) at 0 °C is added triethylamine (278 mL, 2.0 mmol, 2.0 equiv, purchased from Aldrich Chemical Company, Inc., and distilled from CaH<sub>2</sub>) and benzoyl chloride (232 mL, 2.0 mmol, 2.0 equiv, purchased from Aldrich Chemical Company, Inc., and distilled before use). The mixture is stirred at room temperature for



4 h and then EtOAc (50 mL) and H<sub>2</sub>O (10 mL) are added. The aqueous layer is separated and then is extracted with EtOAc (3 x 15 mL). The organic layers are combined, washed with 15 mL of saturated, aqueous sodium bicarbonate solution, dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated under vacuum. The residue is purified by column chromatography (SiO<sub>2</sub>, hexane/*i*-PrOH, 6/1) to give 295 mg (85%) of (*R,R*)-**3** as a white solid. The analytical data for (*R,R*)-**3** are as follows:<sup>4</sup> <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ: 1.76–2.05 (m, 6 H) 2.20–2.28 (m, 2 H), 3.19 (dt, *J* = 10.3, 7.8, 2 H), 3.79 (ddd, *J* = 10.5, 8.8, 5.1, 2 H), 4.59–4.64 (m, 2 H), 7.22–7.36 (m, 6 H), 7.38–7.42 (m, 4 H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ: 24.1, 28.2, 49.1, 58.8, 127.1, 128.2, 129.5, 137.2, 164.5; IR (CHCl<sub>3</sub>) cm<sup>-1</sup>: 2997, 2881, 1624, 1576, 1427, 700; MS (EI, 70 eV): 348, 175, 174, 105; HRMS (*m/z*): Calc. C<sub>22</sub>H<sub>25</sub>N<sub>2</sub>O<sub>2</sub> (M+H): 349.1911; Found: 349.1900; Supercritical Fluid Chromatography: *t*<sub>R</sub> (*R,R*)-**3** 2.86 min (100 %); *t*<sub>R</sub> (*S,S*)-**3** 3.26 min (0 %) (Chiralpak AS, 40 °C, 150 bar, 15 % MeOH in CO<sub>2</sub>, 3.0 mL/min, 220 nm); HPLC: *t*<sub>R</sub> (*S,S*)-**3** 8.2 min (0%); *t*<sub>R</sub> (*R,R*)-**3** 13.2 min (100%) (Chiralpak AD, *i*-PrOH/hexane, 95/5, 0.7 mL/min)

15. The submitters initially used 40 g of KOH pellets for the neutralization. The checkers found that the use of 80 g was necessary to assure extraction of all of the diamine.

16. D-(–)-Tartaric acid (97% GLC) was purchased from Aldrich Chemical Company, Inc., and was used without further purification.

17. The analytical data for (*S,S*)-2,2'-bispyrrolidine•(D)-(–)-tartrate are as follows: mp 214–218 °C; <sup>1</sup>H NMR (500 MHz, D<sub>2</sub>O/DSS) δ: 1.88–1.99

(m, 2 H), 2.08–2.29 (m, 4 H), 2.41–2.49 (br, 2 H), 3.52–3.57 (m, 4 H), 3.94–4.01 (m, 2 H), 4.44 (s, 2 H), 4.84 (br, 6 H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{D}_2\text{O}$ )  $\delta$ : 25.5, 31.0, 49.1, 63.16, 76.6, 181.4; IR (KBr)  $\text{cm}^{-1}$ : 3384, 3242, 2997, 2885, 2717, 2517, 1693, 1610, 1583, 1387, 1124, 1072, 710;  $[\alpha]_{\text{D}}^{24}$   $-17.7$  ( $c = 1.02$ ,  $\text{H}_2\text{O}$ ); Anal. Calcd for  $\text{C}_{12}\text{H}_{22}\text{N}_2\text{O}_6$ : C, 49.65; H, 7.64, N, 9.65. Found: C, 49.98; H, 7.43; N, 9.37. The checkers noted slight chemical shift variations in the NMR spectra of this material, and attribute these to slight differences in concentration and apparent pH in the different samples.

18. The analytical data for (*S,S*)-2,2'-bispyrrolidine are as follows:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$ : 1.31-1.38 (m, 2 H), 1.65-1.84 (m, 6 H), 2.06 (br, 2 H), 2.82-2.97 (m, 6 H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$ : 25.4, 29.0, 46.4, 63.8; IR (KBr)  $\text{cm}^{-1}$ : 3263, 2954, 2867, 2821, 1457, 1442, 1280, 1118, 1076, 892, 869; MS (FAB) ( $m/z$ ) 141; HRMS ( $m/z$ )  $\text{C}_8\text{H}_{17}\text{N}_2$  ( $\text{M}+\text{H}$ ): Calc: 141.1386; Found: 141.1373;  $[\alpha]_{\text{D}}^{24}$  14.82 ( $c = 1.01$ ,  $\text{MeOH}$ ); Anal. Calcd for  $\text{C}_8\text{H}_{16}\text{N}_2$ : C, 68.52; H, 11.50, N, 19.98. Found: C, 68.45; H, 11.64; N, 19.79.

19. For the derivatization procedure see Note 14. The analytical data for (*S,S*)-**3** are as follows:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$ : 1.78–2.05 (m, 6 H), 2.20–2.27 (m, 2 H), 3.22 (dt,  $J = 10.4, 7.8$ , 2 H), 3.80 (ddd,  $J = 10.6, 8.8, 5.1$ , 2 H), 4.60–4.64 (m, 2 H), 7.23–7.33 (m, 6 H), 7.38–7.44 (m, 4 H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$ : 24.3, 28.4, 49.3, 59.0, 127.4, 128.4, 129.8, 137.4, 171.2; MS (EI, 70 eV): 348, 175, 174, 105; HRMS ( $m/z$ ) Calc.  $\text{C}_{22}\text{H}_{25}\text{N}_2\text{O}_2$  ( $\text{M}+\text{H}$ ): 349.1911; Found: 349.1917; Supercritical Fluid Chromatography:  $t_{\text{R}}$  (*R,R*)-**3** 2.86 min (0%);  $t_{\text{R}}$  (*S,S*)-**3** 3.26 min (100%) (Chiralpak AS, 40°C, 150 bar, 15% MeOH in  $\text{CO}_2$ , 3.0 mL/min, 220 nm); HPLC: (*S,S*)-**3**  $t_{\text{R}}$  8.2 min (100%); (*R,R*)-**3**  $t_{\text{R}}$  13.2 min (0%) (Chiralpak AD, *i*-PrOH/hexane, 95/5, 0.7 mL/min)

### Safety and Waste Disposal Information

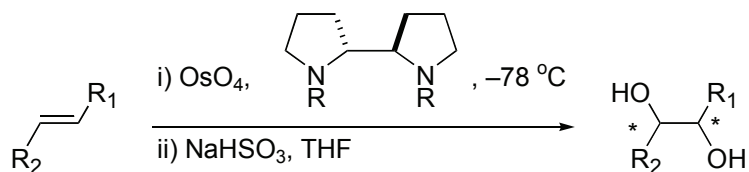
All hazardous materials should be handled and disposed of in accordance with “Prudent Practices in the Laboratory”; National Academy Press; Washington, DC, 1995.

### 3. Discussion

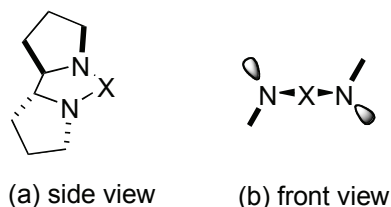
$\text{C}_2$ -Symmetric chiral diamines have found extensive application as additives, auxiliaries and catalysts in asymmetric synthesis.<sup>5</sup> (*R,R*)-2,2'-Bispyrrolidine, initially developed by Hiramama, has been successfully applied



as a ligand for osmium tetroxide in the asymmetric dihydroxylation of olefins<sup>6</sup> (eq 2) and as a ligand in asymmetric hydrogenation.<sup>7</sup>



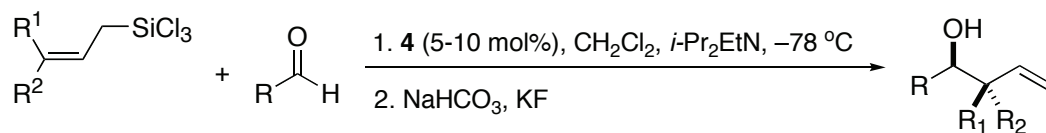
Several syntheses of enantiopure 2,2'-bispyrrolidine have been reported.<sup>4,8,9</sup> The first synthesis described by Masamune and coworkers requires only two steps, but produces a *d,l/meso* mixture of isomers in a sluggish and irreproducible heterogeneous hydrogenation. This short synthesis arrives as the final product by direct resolution of the *d,l/meso* mixture of 2,2'-bispyrrolines.<sup>4</sup> The other routes produce enantiopure 2,2'-bispyrrolines without resolution, but they require multiple-step syntheses from chiral starting materials. For example, the synthesis developed by Kotsuki and coworkers takes 11 steps from (D)-tartaric acid.<sup>8</sup> Most recently, Alexakis reported a five-step synthesis of (*R,R*)-2,2'-bispyrrolidine by asymmetric addition to a chiral imine.<sup>9</sup> In the procedure described herein, the *d,l/meso* mixture of 2,2'-bispyrrolidines is easily synthesized on a large scale by the photodimerization of pyrrolidine developed by Crabtree.<sup>2</sup> The previously reported resolution<sup>4</sup> has been modified such that both enantiomers can be obtained.



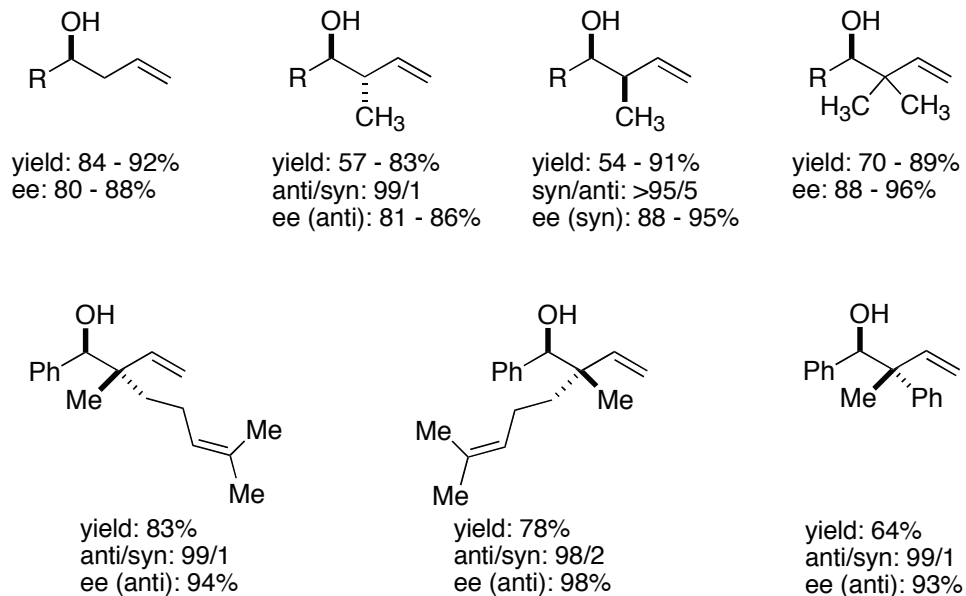
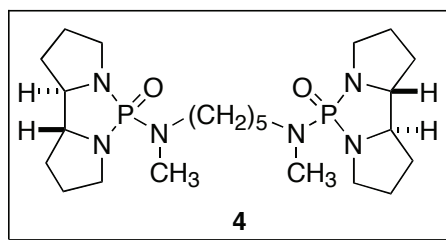
**Figure 2:** Stair-like structure of two pyrrolidine rings

This diamine possesses very interesting structural features that impart useful characteristics as a bidentate ligand. When the two nitrogen atoms function either in a chelate or are covalently bonded to another atom, the two pyrrolidines adopted a stair-like structure, which creates a highly asymmetric environment (Figure 2). This feature was recently exploited in the development of a highly selective catalyst for asymmetric allylations (Figure 3).<sup>3</sup> The addition of allylic trichlorosilanes to unsaturated aldehydes

can be catalyzed by chiral bisphosphoramidate **4** derived from 2,2'-bispyrrolidine to give homoallylic alcohols with excellent diastereo- and enantioselectivities. Of particular note is the catalytic enantioselective construction of quaternary centers by the use of  $\gamma$ -disubstituted allylic silanes. The unique structural features of this diamine together with the ease of preparation bode well for further application in asymmetric synthesis.



R = C<sub>6</sub>H<sub>5</sub>, 4-CH<sub>3</sub>OC<sub>6</sub>H<sub>4</sub>, 2-naphthyl,  
2-furyl, (*E*)-C<sub>6</sub>H<sub>5</sub>CH=CH,  
(*E*)-C<sub>6</sub>H<sub>5</sub>CH=C(CH<sub>3</sub>)



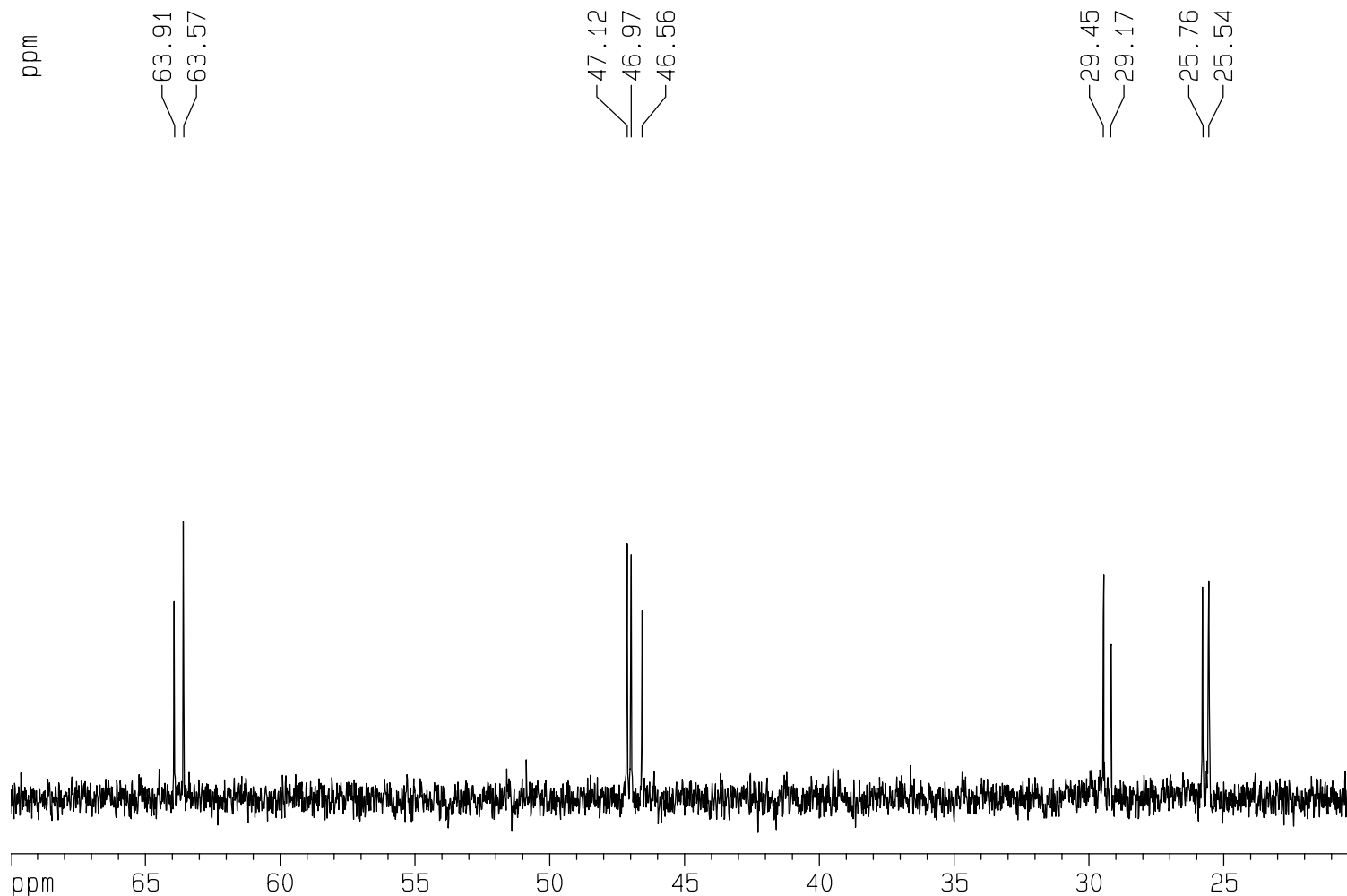
**Figure 3.** Enantioselective addition of allylic trichlorosilanes catalyzed by 2,2'-bispyrrolidine-derived bisphosphoramidate **4**.

1. Department of Chemistry, University of Illinois at Urbana-Champaign, Urbana, IL, 61801.
2. Ferguson, R. R.; Boojamra, C. G.; Brown, S. H.; Crabtree, R. H. *Heterocycles* **1989**, 28, 121.
3. Denmark, S. E.; Fu, J. *J. Am. Chem. Soc.* **2001**, 123, 9488.
4. Oishi, T.; Hiramama, M.; Sita, L. R.; Masamune, S. *Synthesis* **1991**, 789.
5. (a) Bennani, Y. L.; Hannessian, S. *Chem. Rev.* **1997**, 97, 3161. (b) Lucet, D.; Le Gell, T.; Mioskowski, C. *Angew. Chem. Int. Ed.* **1998**, 37, 2580.
6. (a) Hiramama, M.; Oishi, T.; Ito, S. *J. Chem. Soc., Chem. Commun.* **1989**, 665. (b) Oishi, T.; Hiramama, M. *J. Org. Chem.* **1989**, 54, 5834.
7. Hamada, T.; Izawa, K. *Eur. Pat. Appl.* 987271, **2000**.
8. Kotsuki, H.; Kuzume, H.; Gohda, T.; Fukuhara, M.; Ochi, M.; Oishi, T.; Hiramama, M.; Shiro, M. *Tetrahedron: Asymmetry*, **1995**, 6, 2227.
9. Alexakis, A.; Tomassini, A.; Chouillet, C.; Roland, S.; Mangeney, P.; Bernardinelli, G. *Angew. Chem. Int. Ed.* **2000**, 39, 4093.

## Appendix

### Chemical Abstracts Nomenclature; (Registry Number)

- 2,2-Bispyrrolidine; 2,2-Bipyrrrolidine; (74295-58-2)  
 (*R,R*)-2,2'-bispyrrolidine•(L)-tartrate: 2,2'-Bipyrrrolidine, (2*R*,2'*R*)-, (2*R*,3*R*)-  
 2,3-dihydroxybutanedioate (1:1)-; (137037-21-9)  
 (*S,S*)-2,2'-bispyrrolidine•(D)-tartrate: 2,2'-Bipyrrrolidine, [*S*-(*R\**,*R\**)]-, [*S*-  
 (*R\**,*R\**)]-2,3-dihydroxybutanedioate (1:1); (136937-03-6)  
 (*R,R*)-2,2'-Bispyrrolidine: 2,2'-Bipyrrrolidine, (2*R*,2'*R*)-; (137037-20-8)  
 (*S,S*)-2,2'-Bispyrrolidine: 2,2'-Bipyrrrolidine, (2*S*,2'*S*)-; (124779-66-4)



**<sup>13</sup>C-BISPYRROLIDINE PHOTOLYSIS**

**(NMR of photomixture in CDCl<sub>3</sub>)**

**Pyrrolidine - 47.1 ppm**

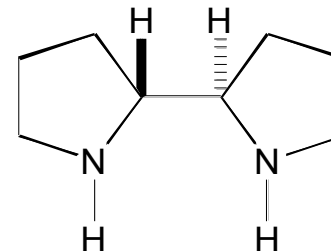
**Meso-Bispyrrolidine - 47.0 ppm**

**(R,R & S,S)-Bispyrrolidine - 46.6 ppm**

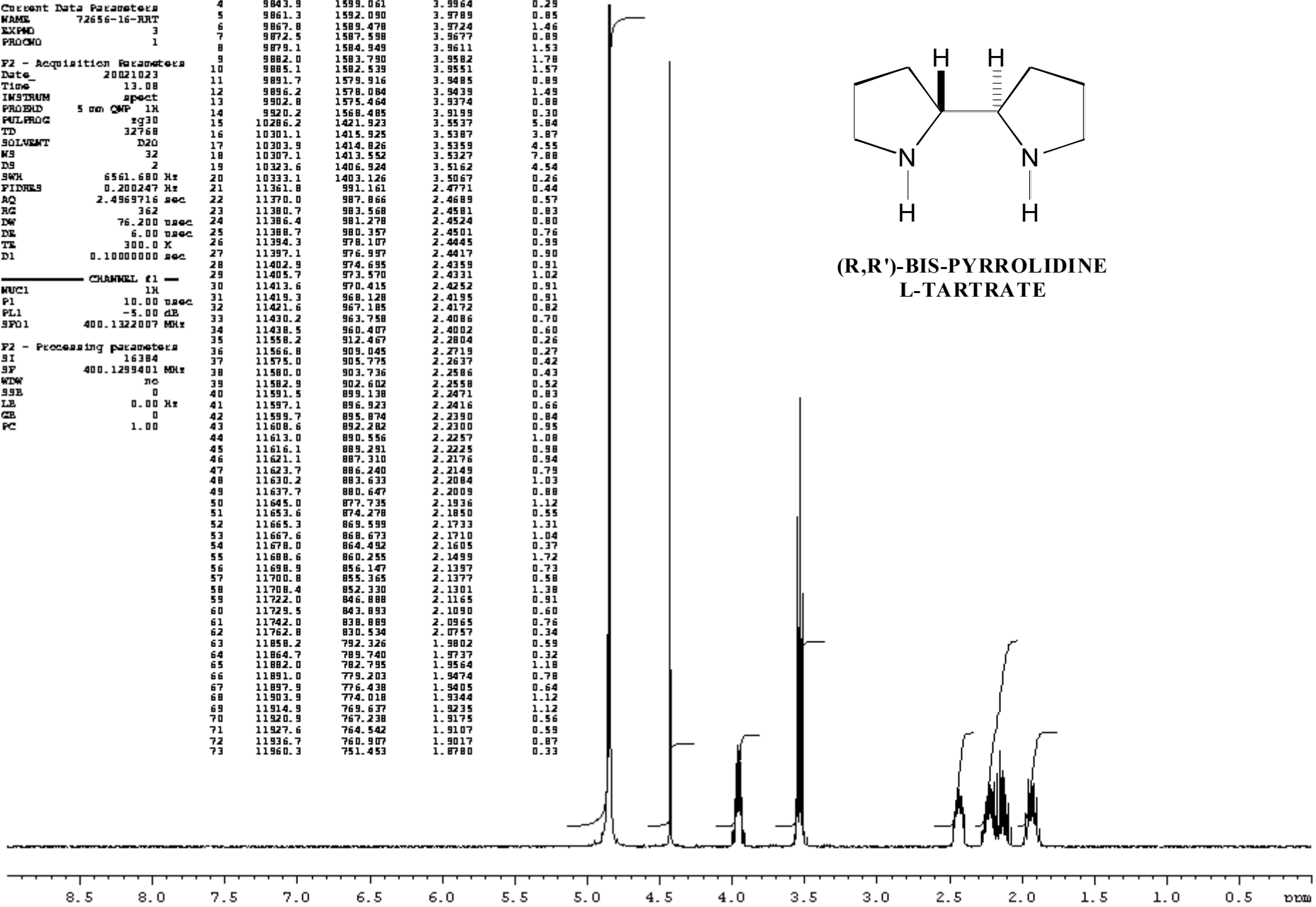
**Peak heights used to calculate approx.  
conversion**

DU-C:/Biokec/ANAL-NMR, USER=lab, NAME=72656-16-RRT, EXPNO=3, PROCNO=1  
 F1=9.000ppm, F2=0.000ppm, MI=0.24cm, MAXI=10000.00cm, PC=1.000

F	ADDRESS	FREQUENCY [Hz]	[PPM]	INTENSITY
1	8943.3	1959.722	4.8977	0.25
2	8992.2	1940.144	4.8488	88.31
3	9411.7	1772.130	4.4289	14.25
4	9843.9	1599.061	3.9964	0.29
5	9861.3	1592.090	3.9789	0.85
6	9867.8	1589.478	3.9724	1.46
7	9872.5	1587.598	3.9677	0.89
8	9879.1	1584.949	3.9611	1.53
9	9882.0	1583.790	3.9582	1.78
10	9885.1	1582.539	3.9551	1.57
11	9891.7	1579.916	3.9485	0.89
12	9896.2	1578.084	3.9439	1.49
13	9902.8	1575.464	3.9374	0.88
14	9920.2	1568.485	3.9199	0.30
15	10286.2	1421.923	3.5537	5.84
16	10301.1	1415.925	3.5387	3.87
17	10303.9	1414.826	3.5359	4.55
18	10307.1	1413.552	3.5327	7.88
19	10323.6	1406.924	3.5162	4.54
20	10333.1	1403.126	3.5067	0.26
21	11361.8	991.161	2.4771	0.44
22	11370.0	987.866	2.4689	0.57
23	11380.7	983.568	2.4581	0.83
24	11386.4	981.278	2.4524	0.80
25	11388.7	980.357	2.4501	0.76
26	11394.3	978.107	2.4445	0.99
27	11397.1	976.997	2.4417	0.90
28	11402.9	974.695	2.4359	0.91
29	11405.7	973.570	2.4331	1.02
30	11413.6	970.415	2.4252	0.91
31	11419.9	968.128	2.4195	0.91
32	11421.6	967.185	2.4172	0.82
33	11430.2	963.758	2.4086	0.70
34	11438.5	960.407	2.4002	0.60
35	11558.2	912.467	2.2804	0.26
36	11566.8	909.045	2.2719	0.27
37	11575.0	905.775	2.2637	0.42
38	11580.0	903.736	2.2586	0.43
39	11582.9	902.602	2.2558	0.52
40	11591.5	899.138	2.2471	0.83
41	11597.1	896.923	2.2416	0.66
42	11599.7	895.874	2.2390	0.84
43	11608.6	892.282	2.2300	0.95
44	11613.0	890.556	2.2257	1.08
45	11616.1	889.291	2.2225	0.98
46	11621.1	887.310	2.2176	0.94
47	11623.7	886.240	2.2149	0.79
48	11630.2	883.633	2.2084	1.03
49	11637.7	880.647	2.2009	0.88
50	11645.0	877.735	2.1936	1.12
51	11653.6	874.278	2.1850	0.55
52	11665.3	868.598	2.1733	1.31
53	11667.6	868.673	2.1710	1.04
54	11678.0	864.492	2.1605	0.37
55	11688.6	860.255	2.1499	1.72
56	11698.9	856.147	2.1397	0.73
57	11700.8	855.365	2.1377	0.58
58	11708.4	852.330	2.1301	1.38
59	11722.0	846.888	2.1165	0.91
60	11729.5	843.893	2.1090	0.60
61	11742.0	838.889	2.0965	0.76
62	11762.8	830.534	2.0757	0.34
63	11858.2	792.326	1.9802	0.59
64	11864.7	789.740	1.9737	0.32
65	11882.0	782.795	1.9564	1.18
66	11891.0	779.203	1.9474	0.78
67	11897.9	776.438	1.9405	0.64
68	11903.9	774.018	1.9344	1.12
69	11914.9	769.637	1.9235	1.12
70	11920.9	767.238	1.9175	0.56
71	11927.6	764.542	1.9107	0.59
72	11936.7	760.907	1.9017	0.87
73	11960.3	751.453	1.8780	0.33



(R,R')-BIS-PYRROLIDINE  
L-TARTRATE



JX-C:/Bruker/MSD-007, USER=Lab, GNAME=72705-5, KEYID=1, F10000-1  
 F1=0.000ppm, F2=0.000ppm, W1=0.50cm, MAXI=10000.00cm, XC=1.000

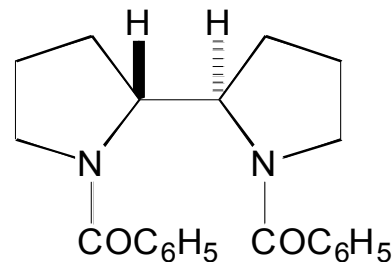
Current Data Parameters  
 NAME 72705-5  
 EXPNO 1  
 PROCNO 1

F2 - Acquisition Parameters  
 Date\_ 20020709  
 Time 14.34  
 INSTRUM spect  
 PROBHD 5 mm QNP 1H  
 PULPROG zg30  
 TD 32768  
 SOLVENT CDCl3  
 NS 16  
 DS 2  
 SWH 6578.947 Hz  
 FIDRES 0.200774 Hz  
 AQ 2.4904180 sec  
 RG 101.6  
 DW 76.000 usec  
 DE 6.00 usec  
 TE 300.0 K  
 D1 0.10000000 sec

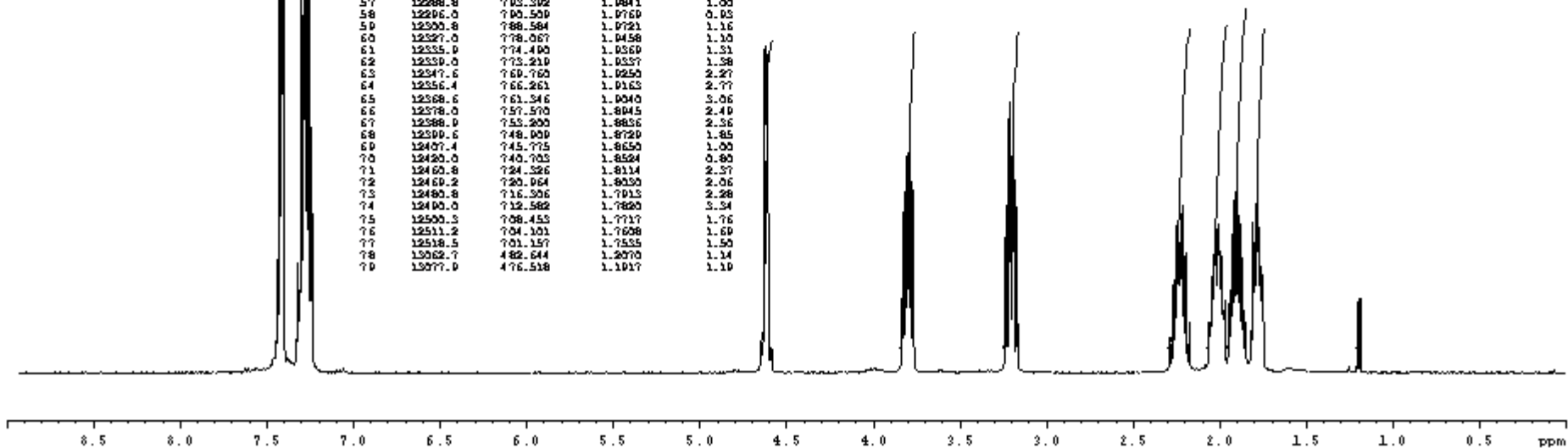
CHANNEL f1  
 NUC1 1H  
 P1 10.50 usec  
 PL1 6.00 dB  
 SFO1 399.8724592 MHz

F2 - Processing parameters  
 SI 16384  
 SF 399.8700207 MHz  
 WDW no  
 SSB 0  
 LB 0.00 Hz  
 GB 0  
 PC 1.00

#	ADDRESS	FREQUENCY [Hz]	[PPM]	INTEGRITY
1	6865.2	2071.203	7.4304	10.00
2	6868.6	2069.867	7.4271	12.23
3	6870.3	2065.551	7.4163	4.76
4	6884.4	2063.509	7.4112	14.68
5	6888.0	2061.781	7.4067	13.62
6	6975.1	2057.896	7.3921	1.73
7	6976.0	2056.576	7.3183	1.32
8	6981.8	2054.411	7.3154	1.22
9	6991.7	2050.416	7.3034	6.60
10	6998.8	2047.579	7.2963	2.67
11	7004.5	2045.299	7.2906	5.02
12	7008.6	2043.651	7.2865	9.18
13	7013.2	2041.800	7.2819	7.32
14	7017.8	2040.037	7.2772	15.00
15	7022.2	2041.165	7.2658	7.42
16	7026.0	2042.639	7.2590	13.23
17	7048.3	2037.700	7.2466	2.18
18	7052.4	2036.065	7.2425	3.97
19	7057.6	2035.070	7.2373	2.62
20	9635.6	1858.789	4.6485	0.43
21	9637.8	1849.862	4.6262	5.19
22	9664.6	1847.129	4.6103	4.45
23	9672.2	1844.098	4.6117	5.04
24	9694.8	1834.097	4.5800	0.40
25	10440.3	1532.032	3.8313	2.59
26	10462.2	1526.873	3.8184	2.88
27	10471.2	1523.241	3.8083	3.44
28	10475.6	1521.492	3.8050	3.42
29	10484.0	1518.083	3.7964	3.32
30	10488.3	1516.339	3.7921	3.52
31	10497.5	1512.677	3.7829	3.13
32	10510.1	1507.696	3.7702	2.00
33	11042.7	1283.760	3.2355	2.68
34	11062.1	1285.076	3.2160	4.29
35	11068.0	1283.233	3.2091	2.87
36	11079.0	1278.811	3.1981	3.28
37	11088.0	1275.537	3.1899	3.83
38	11106.1	1268.306	3.1718	2.63
39	11185.0	1245.018	3.2883	0.55
40	12036.8	906.613	2.2673	1.48
41	12016.5	902.815	2.2578	0.94
42	12025.5	899.126	2.2485	2.33
43	12036.8	894.589	2.2372	2.06
44	12043.7	891.812	2.2303	1.81
45	12056.2	886.777	2.2177	2.67
46	12065.2	883.191	2.2087	1.00
47	12075.1	879.202	2.1987	1.02
48	12096.3	870.698	2.1775	0.80
49	12209.2	825.347	2.0640	0.82
50	12221.0	820.269	2.0513	0.76
51	12229.6	817.171	2.0436	1.52
52	12240.3	812.848	2.0328	1.00
53	12248.2	809.684	2.0249	2.03
54	12261.0	804.566	2.0121	2.37
55	12270.3	800.815	2.0027	1.03
56	12279.6	797.090	1.9934	1.87
57	12288.8	793.392	1.9841	1.00
58	12296.0	790.509	1.9769	0.93
59	12300.8	788.584	1.9721	1.16
60	12327.0	778.067	1.9458	1.10
61	12335.0	774.490	1.9369	1.31
62	12339.0	773.219	1.9337	1.38
63	12347.6	769.760	1.9250	2.27
64	12356.4	766.261	1.9163	2.77
65	12368.6	761.346	1.9040	3.06
66	12378.0	757.570	1.8945	2.49
67	12388.0	753.200	1.8836	2.36
68	12399.6	748.009	1.8729	1.85
69	12407.4	745.775	1.8650	1.00
70	12420.0	740.703	1.8524	0.80
71	12460.8	724.326	1.8114	2.37
72	12469.2	720.064	1.8030	2.06
73	12480.8	716.306	1.7913	2.28
74	12490.0	712.582	1.7820	3.34
75	12500.3	708.453	1.7717	1.76
76	12511.2	704.101	1.7628	1.69
77	12518.5	701.157	1.7535	1.50
78	12562.7	482.644	1.2970	1.14
79	12977.0	476.518	1.1917	1.19

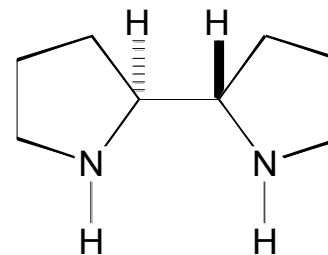


N,N'-BIS-BENZOYL-(R,R')-BIS PYRROLIDINE

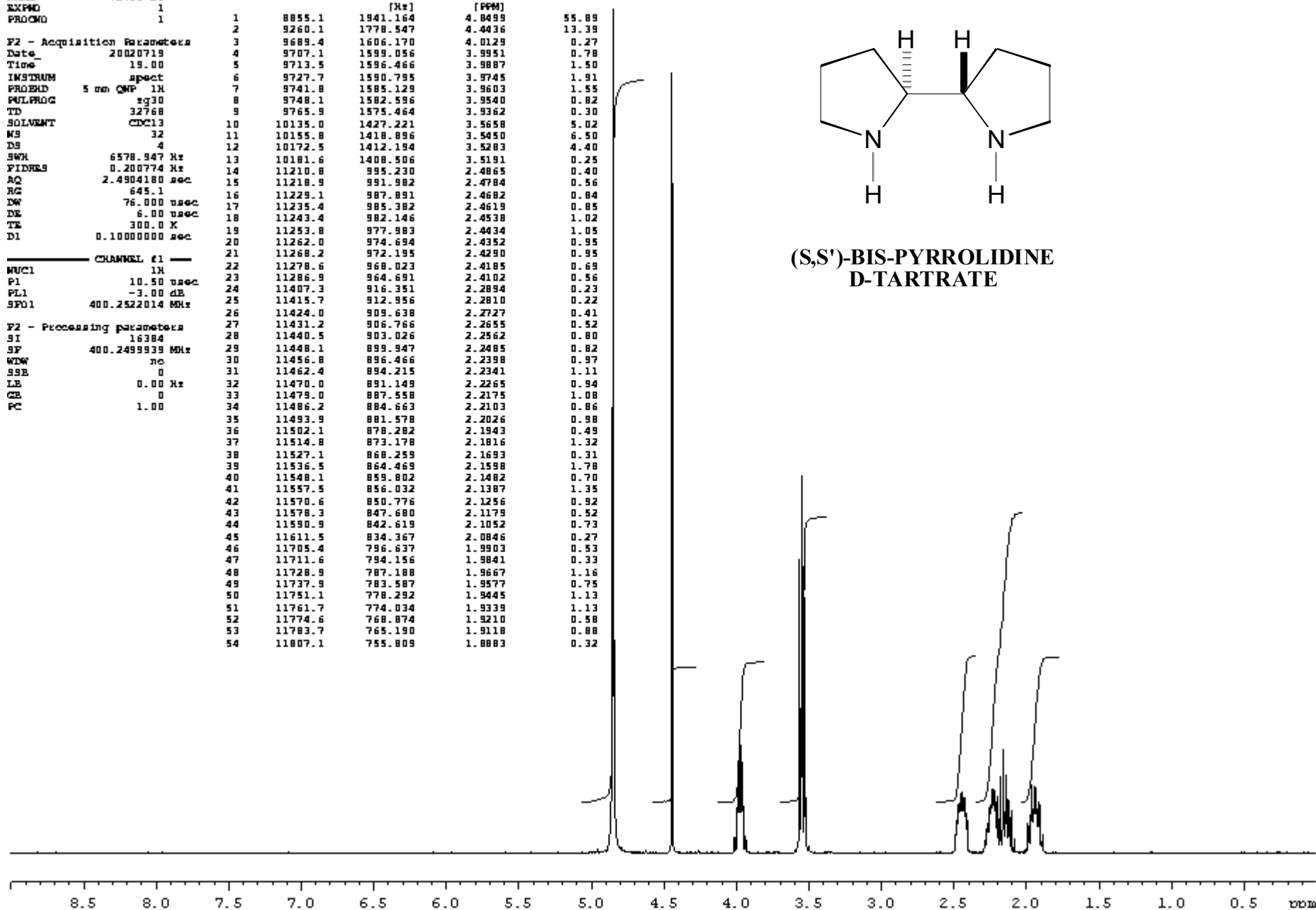


DI=C:/Exeter/XWIN-MMR, USER=lab, NAME=72705-23, EXPNO=1, PROCNO=1  
 F1=9.000ppm, F2=0.000ppm, MI=0.20cm, MAXI=10000.00cm, PC=1.000

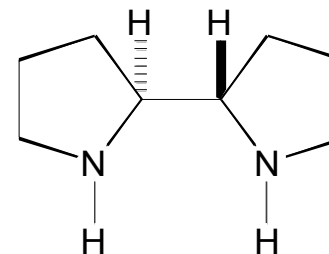
Current Data Parameters	F	ADDRESS	FREQUENCY [Kz]	INTENSITY [PPM]
NAME	72705-23			
EXPNO	1			
PROCNO	1			
F2 - Acquisition Parameters				
Date_	20020719			
Time	19.00			
INSTRUM	spect			
PROBHD	5 mm QNP 1H			
PULPROG	zg30			
TD	32768			
SOLVENT	CDCl3			
NS	32			
DS	4			
SWH	6578.947 Kz			
FIDRES	0.200774 Kz			
AQ	2.4904180 sec			
RG	645.1			
DW	76.000 usec			
DE	6.00 usec			
TE	300.0 K			
D1	0.10000000 sec			
CHANNEL f1				
MUCL	1K			
P1	10.50 usec			
PL1	-3.00 dB			
SFO1	400.2522014 MHz			
F2 - Processing parameters				
SI	16384			
SP	400.2499939 MHz			
WDW	nc			
SSE	0			
LB	0.00 Kz			
GB	0			
PC	1.00			



(S,S')-BIS-PYRROLIDINE  
D-TARTRATE



NU-C2/Benkov/XWIN-NMR, USER=lab, NAME=72705-03, EXPNO=1, PROCNO=1  
 F1=9.000ppm, F2=0.000ppm, MI=0.84cm, MAXI=10000.00cm, PO=1.000



(S,S')-BIS-PYRROLIDINE

F	ADDRESS	FREQUENCY [Hz]	INTEGRITY [PRM]
1	6409.7	2908.911	7.2699
2	10705.8	1188.358	2.9699
3	10720.0	1182.664	2.9557
4	10724.0	1181.067	2.9517
5	10731.1	1178.192	2.9445
6	10737.7	1175.546	2.9379
7	10745.1	1172.592	2.9305
8	10749.2	1170.957	2.9264
9	10762.9	1165.452	2.9127
10	10792.1	1153.777	2.8835
11	10809.1	1146.981	2.8665
12	10820.2	1142.527	2.8554
13	10824.8	1140.671	2.8508
14	10827.9	1139.425	2.8476
15	10836.7	1135.920	2.8389
16	10844.5	1132.785	2.8310
17	10850.4	1130.447	2.8252
18	10853.3	1129.261	2.8222
19	10870.0	1122.583	2.8055
20	11284.9	956.422	2.3903
21	11833.4	736.737	1.8412
22	11838.0	734.889	1.8366
23	11845.6	731.864	1.8291
24	11850.4	729.942	1.8243
25	11855.0	728.081	1.8196
26	11859.3	726.358	1.8153
27	11862.9	724.923	1.8117
28	11867.1	723.246	1.8075
29	11872.1	721.249	1.8025
30	11874.8	720.157	1.7998
31	11878.5	718.269	1.7951
32	11884.9	716.122	1.7897
33	11890.9	713.730	1.7837
34	11896.0	711.681	1.7786
35	11898.8	710.554	1.7758
36	11903.3	708.739	1.7713
37	11907.6	707.045	1.7670
38	11911.1	705.641	1.7635
39	11915.0	704.048	1.7595
40	11923.4	700.688	1.7512
41	11924.7	700.189	1.7499
42	11929.0	698.471	1.7456
43	11934.5	696.254	1.7401
44	11941.2	693.577	1.7334
45	11946.1	691.614	1.7285
46	11953.2	688.781	1.7214
47	11954.9	688.084	1.7197
48	11959.1	686.391	1.7154
49	11960.8	685.704	1.7137
50	11964.1	684.410	1.7105
51	11975.8	679.710	1.6987
52	11980.8	677.718	1.6937
53	11982.8	676.921	1.6918
54	11984.7	676.133	1.6898
55	11987.8	674.923	1.6868
56	11994.9	672.055	1.6796
57	11996.9	671.246	1.6776
58	12002.3	669.093	1.6722
59	12006.5	667.413	1.6680
60	12012.1	665.165	1.6624
61	12016.2	663.551	1.6583
62	12026.3	659.494	1.6482
63	12027.7	658.930	1.6468
64	12033.1	656.782	1.6414
65	12047.0	651.186	1.6274
66	12295.7	551.580	1.3785
67	12300.5	549.656	1.3737
68	12314.2	544.206	1.3601
69	12318.6	542.442	1.3557
70	12325.4	539.695	1.3488
71	12331.0	537.469	1.3432
72	12337.1	534.997	1.3371
73	12343.9	532.281	1.3303
74	12348.4	530.470	1.3257
75	12354.9	527.869	1.3192
76	12362.2	524.943	1.3119
77	12366.1	523.412	1.3081
78	12385.9	515.465	1.2882

