

A Publication of Reliable Methods for the Preparation of Organic Compounds

Working with Hazardous Chemicals

The procedures in *Organic Syntheses* are intended for use only by persons with proper training in experimental organic chemistry. All hazardous materials should be handled using the standard procedures for work with chemicals described in references such as "Prudent Practices in the Laboratory" (The National Academies Press, Washington, D.C., 2011; the full accessed of charge text can be free at http://www.nap.edu/catalog.php?record_id=12654). All chemical waste should be disposed of in accordance with local regulations. For general guidelines for the management of chemical waste, see Chapter 8 of Prudent Practices.

In some articles in *Organic Syntheses*, chemical-specific hazards are highlighted in red "Caution Notes" within a procedure. It is important to recognize that the absence of a caution note does not imply that no significant hazards are associated with the chemicals involved in that procedure. Prior to performing a reaction, a thorough risk assessment should be carried out that includes a review of the potential hazards associated with each chemical and experimental operation on the scale that is planned for the procedure. Guidelines for carrying out a risk assessment and for analyzing the hazards associated with chemicals can be found in Chapter 4 of Prudent Practices.

The procedures described in *Organic Syntheses* are provided as published and are conducted at one's own risk. *Organic Syntheses, Inc.,* its Editors, and its Board of Directors do not warrant or guarantee the safety of individuals using these procedures and hereby disclaim any liability for any injuries or damages claimed to have resulted from or related in any way to the procedures herein.

September 2014: The paragraphs above replace the section "Handling and Disposal of Hazardous Chemicals" in the originally published version of this article. The statements above do not supersede any specific hazard caution notes and safety instructions included in the procedure.





Submitted by Gene W. Wong, Tyler T. Adint, Clark R. Landis.¹ Checked by Neil Strotman, James Cuff and David Hughes.

1. Procedure

Caution! Carbon monoxide is a highly toxic gas and manipulations should be conducted in a well-ventilated fume hood in the vicinity of a carbon monoxide detector. Hydrogen gas is highly flammable and explosive gas. Precautions should be taken when using synthesis gas (H_2 /CO mixtures).

A. Allyl (t-butyldimethyl)silyl ether 1. An oven-dried three-necked, 500mL round-bottomed flask equipped with a 3-cm PTFE-coated oval stir bar is charged with allyl alcohol (8.6 g, 0.15 mol, 1.0 equiv), imidazole (21.1 g, 0.31 mol, 2.1 equiv), and dimethylformamide (100 mL). (Note 1) The flask is fitted with a gas inlet adapter connected to a nitrogen line and a gas bubbler. The other two necks are capped with rubber septa; a thermocouple probe is inserted through one of the septa. (Note 2) The stirred solution is cooled to 4 °C with an ice-water bath, then *t*-butyldimethylsilyl chloride (25.5 g, 0.17 mol, 1.1 equiv) is added in portions over 2 min. (Note 3) The water bath is removed and the hazy solution is stirred 15 h at 22–23 °C. (Note 4) The reaction mixture is transferred to a 500-mL separatory funnel along with hexanes (200 mL) and water (60 mL). After mixing and settling, the aq. layer is removed and the organic layer washed with brine (75 mL). The hazy organic layer is filtered through a bed of sodium sulfate (50 g) in a medium porosity sintered glass funnel into a 500-mL round-bottomed flask, using hexanes (2 x 50 mL) to rinse the filter cake. The filtrate is concentrated by rotary evaporation (40 °C, 20 mmHg) to an oil (22 g). Since a substantial portion of the product co-distills during concentration, the distillate from the receiver flask is re-concentrated, providing additional product (2.8 g). The combined crude product is purified by column chromatography (Note 5) to afford allyl (*t*-butyldimethyl)silyl ether **1** (18.1–18.5 g, 71–73% yield) as a colorless oil. (Note 6)

B. (2*R*)-3-[[(1,1-dimethylethyl)dimethylsilyl]oxy]-2-methylpropanal 2. In a glovebox, bis[(S,S,S)-DiazaPhos-SPE] (9.2 mg, 0.0070 mmol, 0.024 mol%) is added to a 20 mL vial followed by 0.22 mL CDCl₃. To a separate 4 mL vial is added Rh(acac)(CO)₂ (10 mg) followed by 1.94 mL toluene to prepare a 20 mM stock solution; 0.29 mL (1.5 mg, 0.0058 mmol, 0.020 mol%) of this stock solution is transferred to the 20 mL vial containing the ligand solution (Notes 7, 8). Allyl (t-butyldimethyl)silyl ether 1 (5.00 g, 6.17 mL, 29 mmol, 1.0 equiv) is added to the vial. The solution is divided into 4 x 4 mL vials, each containing a 5-mm glass bead, and loaded into a Symyx Heated Orbital Shaker System (HOSS) contained within the glove box. (Note 9) The system is taken through 3 cycles of pressurization (150 psig of 1:1 H₂:CO)/depressurization (0 psig) to replace the nitrogen atmosphere with synthesis gas. The system is pressurized with 150 psig 1:1 H₂:CO and heated to 60 °C for 16 h. (Note 10) The mixture is cooled and the system depressurized, then the vials are removed from the glove box. (Note 15) The hydroformylation reaction mixture is purified immediately (Note 16) by flash column chromatography (Note 17) to afford colorless oils of branched isomer (R)-2 (3.24-3.43 g, 55-58% yield, Notes 18-20) and linear isomer **3** (1.62–1.70 g, 27–29% yield, Note 21).

2. Notes

1. The following reagents and solvents in step A were used as received: allyl alcohol (Sigma-Aldrich), imidazole (Acros, 99%), *t*-butyldimethylsilyl chloride (Acros, 98%), anhydrous DMF (Sigma-Aldrich, 99.8%), hexanes (Fisher, ACS reagent, >98.5%), ethyl acetate (Fisher, ACS reagent, >99.5%) and silica gel (Fisher, 230-400 mesh, 60 Å).

2. The internal temperature is monitored using a J-Kem Gemini digital thermometer with a Teflon-coated T-Type thermocouple probe (12-inch length, 1/8 inch outer diameter, temperature range -200 to +250 °C).

3. The reaction warmed to 25 °C over 5 min after addition of TBS-Cl.

4. The reaction was monitored by TLC, 5% EtOAc/hexanes, $R_{\rm f}$ 0.6, $KMnO_4\,stain.$

5. Silica gel (250 g) was slurry-packed in a 5-cm diameter column using 2.5% EtOAc/hexanes. The product was eluted with 2.5% EtOAc/hexanes, collecting 100 mL fractions. Fractions 5-13 were combined and concentrated by rotary evaporation (40 °C, 20 mmHg) to afford 1 (14.5–15.4 g) as a colorless oil. The distillate from concentration was re-concentrated to provide an additional 2.7–4.0 g (combined yield, 18.1–18.5 g, 71–73%). The distillate from the final product concentration was assayed by ¹H NMR using toluene as an internal standard, indicating 2.4 g (10% yield) 1 was present in the distillate.

6. *Allyl (t-butyldimethyl)silyl ether 1* has the following physical and spectroscopic data: ¹H NMR (400 MHz, CDCl₃) δ : 0.09 (s, 6 H), 0.93 (s, 9 H), 4.18–4.20 (m, 2 H), 5.07–5.11 (m, 1 H), 5.25–5.30 (m, 1 H), 5.89–5.97 (m, 1 H); ¹³C NMR (100 MHz, CDCl₃) δ : –5.0, 18.6, 26.2, 64.3, 114.1, 137.8; GC-MS (EI) *m/z*: 172 (6 %) [M⁺], 157 (6%), [M - CH₃], 116 (31%), 115 (100%) [M - *t*-Bu], 99 (21%), 85 (69%) [M – Me₂,*t*-Bu], 75 (28%), 59 (48%); GC purity: 98% (t_R = 4.4 min; conditions: Agilent DB35MS column; 30 m x 0.25 mm; initial temp 60 °C, ramp at 20 °C/min to 280 °C, hold 15 min).

7. The following reagents and solvents in step B were used as received by the checkers: dicarbonylacetylacetonato rhodium(I) (Strem), toluene (Sigma Aldrich, anhydrous, >99.9%), CDCl₃ (Sigma-Aldrich, 99.8% atom % D), SynGas (49% carbon monoxide/51% hydrogen, Airgas) and bis[(S,S,S)-DiazaPhos-SPE] ligand: 2,2',2'',2'''-(1,2-phenylenebis[(1S,3S)tetrahydro-5,8-dioxo-1*H*-[1,2,4]diazaphospholo[1,2-a]pyridazine-2,1,3(3*H*)triyl])tetrakis(N-[(1S)-1-phenylethyl])benzamide (Sigma-Aldrich). The ligand was prepared by the submitters according to their published procedure.² The submitters recrystallized dicarbonylacetylacetonato rhodium(I) from toluene and hexanes as fine green crystals. CHCl₃ may be used instead of CDCl₃.

8. Accurate volumes were measured and transferred using an Eppendoft® pipette.

9. Symyx HOSS equipment is described on the Symyx website. http://symyx.desantisbreindel.com/page.php?id=71

10. The submitter's equipment and experimental protocol are outlined in this Note and in Notes 11-14. A heavy wall reaction tube (Ace Glass #15 Ace-Tread®, 30 cm length x 38.1 mm O.D., 185 mL capacity) and a 0.5 x 0.125 inch magnetic stir bar are dried in a 125 °C oven overnight. In a glove box, the reaction tube is charged with stock solutions of Rh(acac)(CO)₂ and Bis[(*S*,*S*,*S*)-DiazaPhos-SPE] using a 1000 μ L Eppendoft® pipette followed by 5 grams of substrate. The reaction tube is attached to the reactor head (Note 11). Notably, the addition of the alkene to the catalyst solution resulted in a yellow-white suspension due to partial precipitation of ligand and/or catalyst-ligand complex.

A blast shield must be used whenever the reactor is pressurized and safety procedures for using pressure tubes described in the Ace-Glass® catalog should be reviewed and followed.

The assembled reactor is removed from the glove box, placed in a fume hood, connected to the synthesis gas source and taken through 5 cycles of pressurization (150 psig of 1:1 H₂:CO)/depressurization (0 psig) to replace the nitrogen atmosphere with synthesis gas (Notes 12, 13, Figure 2). The reactor is then submerged in a heated silicon oil bath at the desired temperature. As synthesis gas is consumed, the reactor is repressurized to 150 psi to maintain approximately constant pressure (Note 14). After 2–3 hours (~30–40 psi of synthesis gas consumed) the suspension transforms to a homogeneous yellow solution. In six hours, ~90 psi of synthesis gas is consumed. At the end of the reaction time, the reactor is depressurized.

11. A custom-made reactor head used for hydroformylations is shown in Figure 1. The following parts were used to assemble the reactor head: **a**, Alltech® septum (High-temp, 3/8 in., AT79231) for aliquot-abstractions using a gas-tight syringe, **b**, Swagelok® Brass 1-Piece 40 Series Ball Valve (1.6 Cv, 1/4 in. MNPT x 1/4 in. Swagelok Tube Fitting; product #: B-43M4-

S4), c, Swagelok® Brass Pipe Fitting, Cross (1/4 in. Female NPT; product #: B-4-CS), d, Brass Pipe Fitting, Hex Nipple (1/4 in. Male NPT), e, Swagelok® Brass Pipe Fitting, Elbow (1/4 in. Female NPT; product #: B-4-E), f, Ashcroft® 0-160 psig pressure gauge (1/4 in. NPT, 3.5 in. Dial; McMaster-Carr 3846K311 0-160 psig range), g, Brass Pipe Fitting, Close Nipple (1/4 in. Male NPT), h, #15 Ace-Thred® (15 mm thread, 1/4 in. NPT) PTFE Swagelok adapter; Prod. #: 5844-74), i, Kalrez® 6375 O-ring (9.30) mm x 2.40 mm Part #: K31016K6375), j, #15 Ace Glass® pressure tube (30.5 cm L, 38.1 mm OD, Prod. #: 8648-33), k, Swagelok® Brass 1-Piece 40 Series 3-Way Ball Valve (0.75 Cv, 1/4 in. FNPT; product #: B-43XF4), I, Brass Pipe Fitting (1/4 in. male NPT to 1/4 in. male Swagelok Tube Fitting), m, SS tubing (1/4 in OD, 2 1/2 in. length), and n, Swagelok \mathbb{R} SS Instrumentation Quick-Connect Stem w/ Valve, (0.2 Cv, 1/4 in. Swagelok Tube Fitting, Part #: SS-QC4-D-400). Threads **b**, **d**, **f**, **g**, and **l** were wrapped with PTFE tape prior to assembly. A thorough pressure check of reactor should be taken before conducting an experiment. The most common source of a leak is between the brass pipe fitting \mathbf{g} and the plastic #15 Ace-Thred adapter **h**. Once assembled with the 185-mL pressure tube, the reactor is rather cumbersome to transport—the use of an 11.5" (W) x 13.5" (L) x 5.25" (D) Rubbermaid® dishpan with a 3"(D) x 1" (W) rectangle cut in the tub on the width side was used to partially hold the reactor.

12. A reverse-threaded regulator is connected to a synthesis gas cylinder and Swagelok® Quick-Connects are used to attach to the reactor manifold. The synthesis gas cylinder was obtained from AirGas Inc. as a custom mixture ($48.3\pm2\%$ carbon monoxide balanced with hydrogen gas).

13. The reactor has two possible points of entry: Swagelok® Ball valve **b** fitted with a GC septum, for gas-tight syringe aliquots, and the Swagelok® 3-way Ball Valve **k**, for pressurizing and depressurizing the reactor. In Figure 2, **k** is opened carefully to the synthesis gas cylinder, charging the apparatus to 150 psig (it is advisable to set the regulator on the cylinder to ca. 150 psig and to have a safety shield in place). The valve on **k** is then opened to vent, releasing synthesis gas from the apparatus. After the pressure is reduced to <40 psi, the valve is turned back to the original closed position constituting one cycle. This procedure is repeated for five cycles and the reactor pressure is set at 150 psi. The glass tube of the reactor is lowered into the oil bath for hydroformylation as seen in the far-right picture.

14. Synthesis gas is added manually to maintain at least 100 psig reactor pressure. It is not advisable to maintain reactor pressure by keeping the reactor open to the regulator on the synthesis gas cylinder because, in the event of a leak on the reactor or supply lines, large amounts of H_2 and CO could be released. A carbon monoxide detector is installed near the gas cylinder. Commonly, the synthesis gas line is detached from the reactor at the Swagelok® Quick-Connect during reaction and reconnected when adding more gas. However, if the synthesis gas line is not needed for other reactions, the Swagelok® Quick-Connect system can remain assembled throughout the reaction.

15. ¹H NMR of the crude product mixture indicated >99% conversion of alkene and a branched: linear (2:3) ratio of 2:1.

16. Aldehyde **2** is air-sensitive and flash chromatography should be performed immediately after depressurizing the reactor and the purified product stored in a freezer.

17. Silica gel (250 g) was slurry-packed in a 5-cm diameter column using 5% EtOAc/hexanes. The product was eluted with 5% EtOAc/hexanes, collecting 50 mL fractions, monitored by TLC. (10% EtOAc/hexanes, $R_f 1 = 0.7$, (*R*)-2 = 0.38, 3 = 0.31, visualized with potassium permanganate stain, prepared as follows: 3 g KMnO₄, 20 g potassium carbonate, 5 mL of a 5% (w/w) solution of aqueous sodium hydroxide, and 300 mL of deionized water.) Fractions 9-26 were combined and concentrated by rotary evaporation (40 °C, 20 mmHg) to afford 2 (3.24–3.43 g) as a colorless oil. Fractions 29-38 were combined and concentrated by rotary evaporation (40 °C, 20 mmHg) to afford 3 (1.62–1.70 g) as a colorless oil.

18. (2R)-3-[[(1,1-Dimethylethyl)dimethylsilyl]oxy]-2-methylpropanal **2** has the following physical and spectroscopic data: $[\alpha]_D^{25}$ -34 (c 1.0, CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃) δ : 0.06 (s, -Si(CH₃)₂C(CH₃)₃, 6 H), 0.89 (s, -Si(CH₃)₂C(CH₃)₃, 9 H), 1.10 (d, J = 6.9 Hz, -CHCH₃, 3 H), 2.52– 2.56 (m, -CHCH₃, 1 H), 3.82 (dd, J = 6.4, 10.2 Hz, -CH₂OSi, 1 H), 3.86 (dd, J = 5.2 Hz, 10.2, -CH₂OSi, 1 H), 9.74 (d, J = 1.6 Hz, CHO-CH, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ : -5.33, -5.31, 10.5, 18.4, 26.0, 49.0, 63.7, 204.9; IR (neat): 2957, 2931, 2859, 1736 (C=O), 1473, 1258, 1101, 1033, 838, 778 cm⁻¹; GC-MS *m/z* (relative intensity): 145 (100) [M - *t*-Bu], 115 (95) [SiMe₂*t*-Bu], 101 (31), 85 (25) [Si*t*-Bu], 75 (54), 59 (25); GC purity: 98% (t_R = 7.4 min, same conditions as in note 6); ee 94–96% determined by SFC analysis of benzylamine reductive amination derivative as described in Note 20. The aldehyde oxidizes at a rate of about 1% per week when stored in a - 20 °C freezer.

19. The (S)-enantiomer of **2** was prepared by the same procedure using Bis[(R,R,S)-DiazaPhos-SPE] as ligand; $[\alpha]_D^{25}$ +33 (c 1.0, CH₂Cl₂); ee 88%.

20. The submitters determined chiral purity by gas chromatographic analysis on a Varian Chrompack system using a β -DEX 225 capillary column from Supelco, 30 m x 0.25 mm ID x 0.25 um film thickness. The analytical method used to resolve the enantiomers as follow: 65 °C hold for 70 min, t_R(*R*)-2: 60.8 min, t_R(*S*)-2: 62.4 min. The checkers determined chiral purity by formation of the reductive amination product with benzylamine and analysis by supercritical fluid chromatography (SFC): tandem columns: 25 cm OZ : 25cm OZ, isocratic 8% 25mM *i*-butylamine in 2-propanol, 100 bar, 2.0 mL/min for 18 min. t_R(*R*)-4 = 11.5 min, t_R(*S*)-4 = 14 min.

Procedure for the preparation of the reductive amination product with benzylamine follows.



To a 20-mL vial equipped a 0.7 cm stir bar is added sequentially sodium triacetoxyborohydride (235 mg, 1.1 mmol), chloroform (2 mL), aldehyde 2 (73 mg, 0.36 mmol), and benzylamine (37 mg, 0.34 mmol). The heterogeneous mixture is stirred 16 h at ambient temperature, then quenched with 5 mL sat. NaHCO₃, stirring the biphasic mixture for 5 min. Dichloromethane (10 mL) is added, the layers are separated, and the organic layer dried by filtering through 2 g of sodium sulfate. The filtrate is concentrated by rotary evaporation (40 °C, 20 mmHg) to afford crude 4 (115 mg). The product is purified by silica gel chromatography using 10 g silica with an eluent of 97:2:0.5 CH₂Cl₂:MeOH:Et₃N, collecting 10 mL Fractions 7-10 were combined and concentrated by rotary fractions. evaporation to provide product 4 (68 mg, 67% yield) as a colorless oil having the following physical and spectroscopic data: TLC: $R_f = 0.1$ (97:2:0.5 CH₂Cl₂:MeOH:Et₃N); ¹H NMR (500 MHz, CDCl₃) δ: 0.04 (s, 6 H), 0.89 (s, 9 H), 0.91 (d, J = 6.7 Hz, 3 H), 1.68 (br s, 1 H), 1.85–1.91 (m, 1 H), 2.51 (dd, J = 6.0, 11.6 Hz, 1 H), 2.68 (dd, J = 6.8, 11.6 Hz, 1 H), 3.50-3.57 (m, 2 H), 3.77–3.82 (m, 2 H), 7.25–7.26 (m, 1 H), 7.31–7.33 (m, 4 H); ¹³C NMR (125 MHz, CDCl₃) δ: -5.24, -5.21, 15.6, 18.5, 26.1, 36.1, 53.6, 54.5, 127.0, 128.3, 128.5, 140.9.

21. Linear product **3** has the following spectroscopic data: ¹H NMR (400 MHz, CDCl₃) δ : 0.04 (s, -Si(CH₃)₂C(CH₃)₃, 6 H), 0.88 (s, -Si(CH₃)₂C(CH₃)₃, 9 H), 1.86 (tt, J = 6.0, 7.1 Hz -CH₂CH₂CH₂-, 2 H), 2.50 (dt, J = 7.1, 1.8 Hz, CH₂CHO, 2 H), 3.65 (t, J = 6.0 Hz, -CH₂OSi, 3 H), 9.79 (t, J = 1.7 Hz, CHO, 1 H); ¹³C NMR (100 MHz, CDCl₃) δ : -5.2, 18.5, 25.7, 26.1, 41.0, 62.3, 202.9; GC purity: 96% (t_R = 7.9 min, same conditions as in Note 6)

Safety and Waste Disposal Information

All hazardous materials should be handled and disposed of in accordance with "Prudent Practices in the Laboratory"; National Academies Press; Washington, DC, 2011.

3. Discussion

Protected "Roche Aldehydes" (e.g., 2 (2R)-3-[[(1,1dimethylethyl)dimethylsilyl]oxy]-2-methylpropanal) are common starting materials for the synthesis of polyketides and related molecules.^{3,4} Compared with the common reduction-to-alcohol-followed-by-selective-oxidation-toaldehyde route to 2 from "Roche Ester",⁴ hydroformylation of the protected commodity monomer, allyl alcohol, provides Roche Aldehyde derivatives rapidly, at low cost, and in an easily scalable process. For comparison purposes we have collected the following approximate costs of substrates, normalized to 25 g units, from a common supplier: Roche ester (\$350/25g), allyl alcohol (\$1.00/25g). The only byproducts of the enantioselective hydroformylation of **1** is the corresponding linear aldehyde; although achiral the linear aldehyde is isolated cleanly and constitutes a useful synthetic material also. On larger scales, it should be possible to separate the linear and branched aldehydes by careful vacuum distillation; we have not yet optimized the distillation conditions. An advantage of hydroformylation routes to chiral aldehydes is the absence of acids or bases in the reaction solution that catalyze racemization and condensation reactions. We note that although the Roche Ester has been synthesized by asymmetric hydrogenation of the methyl 2-(hydroxymethyl)-prop-2-enoate,⁵ there is no report of a catalytic hydrogenation route to enantiopure Roche Aldehyde.

Figure 1. The submitters assembled reactor with parts indicated



Figure 2. The submitters' reactor in-use



- Department of Chemistry, University of Wisconsin, Madison, Wisconsin, 53706-1322. Email: Landis@chem.wisc.edu. We thank The Dow Chemical Company for their donation of Rh(acac)(CO)₂. Funding was provided by the National Science Foundation (CHE-0715491 and a graduate fellowship for G.W.W.).
- Clark, T. P.; Landis, C. R.; Freed, S. L.; Klosin, K.; Abboud, K. A. J. Am. Chem. Soc. 2005, 127, 5040–5042.
- **3.** For a general review on polyketides stereotetrads, see: Koskinen, A. M. P.; Karisalmi, K. *Chem. Soc. Rev.* **2005**, *34*, 677–690.
- Roche Aldehyde for use in natural product syntheses see: (a) Früstner, A.; Nevado, C.; Waser, M.; Tremblay, M.; Chevrier, C.; Teplý, F.; Aïssa, C.; Moulin, E.; Müller, O. J. Am. Chem. Soc. 2007, 129, 9150– 9161; (b) Canova, S.; Bellosta, V.; Bigot, A.; Mailliet, P.; Mignani, S.; Cossy, J. Org. Lett. 2007, 9, 145–148; (c) Ferrié, L.; Reymond, S.; Capdevielle, P.; Cossy, J. Org. Lett. 2006, 8, 3441–3443; (d) Ehrlich, G.; Kalesse, M. Synlett 2005, 4, 655–657; (e) Smith, A. B., III, Brandt, B. M. Org. Lett. 2001, 3, 1685–1688.
- (a) Qiu, M.; Wang, D.-Y.; Hu, X.-P.; Huang, J.-D.; Yu, S.-B.; Deng, J.; Duan, Z.-C.; Zheng, Z. *Tetrahedron: Asymmetry* 2009, 20, 210–213. (b) Pautigny, C.; Jeulin, S.; Ayad, T.; Zhang, Z.; Genêt, J.-P.; Ratovelomanana-Vidal, V. *Adv. Synth. Catal.* 2008, 350, 2525–2532. (c) Holz, J.; Schäffner, B.; Zayas, O.; Spannenberg, A.; Börner A. *Adv.*

Synth. Catal. **2008**, *350*, 2533–2545. (d) Wassenaar, J.; Kuil, M.; Reek J. N. H. *Adv. Synth. Catal.* **2008**, *350*, 1610–1614. (e) Jeulin, S.; Ayad, T.; Ratovelomanana-Vidal, V.; Genêt, J.-P. *Adv. Synth. Catal.* **2007**, *349*, 1592–1596. (f) Shimizu, H.; Saito, T.; Kumobayashi, H. *Adv. Synth. Catal.* **2003**, *345*, 185–189.

Appendix Chemical Abstracts Nomenclature; (Registry Number)

Silane, (1,1-dimethylethyl)dimethyl(2-propen-1-yloxy)-; (85807-85-8) Silane, chloro(1,1-dimethylethyl)dimethyl-; (18162-48-6)

Prop-2-en-1-ol; (107-18-6)

Rhodium, dicarbonyl(2,4-pentanedionato-κ-O2,κ-O4)-, (SP-4-2)-; (14874-82-9)

Benzamide, 2,2',2",2"'-[1,2-phenylenebis[(1*S*,3*S*)-tetrahydro-5,8-dioxo-1H-[1,2,4]diazaphospholo[1,2-a]pyridazine-2,1,3(3*H*)-triyl]]tetrakis[*N*-[(1*S*)-1-phenylethyl]-; (851770-14-4)

Propanal, 3-[[(1,1-dimethylethyl)dimethylsilyl]oxy]-2-methyl-, (2*R*)-; (97826-89-6),

Butanal, 4-[[(1,1-dimethylethyl)dimethylsilyl]oxy]-; (87184-81-4)



Clark R. Landis was born in Aurora, IL in 1956. After completing his Ph.D. at the University of Chicago in 1983 under the direction of Jack Halpern, Professor Landis held professional positions at the Monsanto Company Corporate Research Lab, the University of Colorado-Boulder, and, since 1990, the University of Wisconsin-Madison. His research interests include bonding theory, computational methods, instrumentation development, chiral ligand synthesis, enantioselective catalysis, catalytic alkene polymerization, and the mechanisms of catalytic reactions.



Gene W. Wong was born in Reno, Nevada in 1985. He received his undergraduate chemistry degree from University of Nevada-Reno, where he conducted research with Prof. Brian J. Frost. He then moved to University of Wisconsin-Madison, where he is currently pursuing a Ph.D. in the research group of Prof. Clark R. Landis as a NSF Predoctoral Fellow. His research focuses on the development of bisdiazaphospholane libraries and rhodium catalyzed hydroformylation.



Tyler T. Adint was born in Fairbanks, Alaska in 1984. He received his undergraduate degree in 2003 from Lewis & Clark College, where he conducted research with Prof. Louis Y. Kuo. He is now pursuing his Ph.D. at the University of Wisconsin-Madison in the research group of Prof. Clark R. Landis. His current research interests concern the synthesis of bisdiazaphospholane ligands that utilize secondary interactions to control selectivity in rhodium catalyzed hydroformylations.

		Peak	?(F1)	[ppm] ?(F1)	[Hz]	Intensity
Current D	ata Parameters	1	7.2708	2907.3749	0.18	-
NAME	32077-152	2	5.9655	2385.4245	0.29	
EXPNO	2	3	5.9509	2379.5864	0.13	
PROCNO	1	4	5.9394	2374.9879	0.33	
1110 0110	-	5	5.9343	2372.9486	0.16	
F2 - Acqu	isition Parameters	6	5.9282	2370.5094	0.14	
Date	20110409	7	5.9225	2368.2301	0.28	
Time	13.49	8	5.9113	2363.7516	0.18	
INSTRUM	spect	9	5.9084	2362.5920	0.18	
PROBHD	5 mm ONP 1H/1	10	5.8965	2357.8335	0.26	
PULPROG	za30	11	5 8852	2353 3150	0 18	
TD	32768	12	5 3057	2121 5903	0 18	
SOLVENT	CDC13	13	5 3007	2119 5910	0.10	
NS	32	14	5 2991	2118 9512	0.30	
DS	2	15	5 2958	2117 6316	0.23	
SWH	6578.947 Hz	16	5 2012	2115 7022	0.52	
FIDRES	0.200774 Hz	17	5 2626	2104 2550	0.10	
AO	2.4904180 sec	10	5.2020	2104.3333	0.13	
RĜ	71.8	10	5.2579	2102.4703	0.35	
DW	76.000 usec	19	5.2334	2100.0771	0.42	
DE	7.00 usec	20	J.2485	2098.7177	0.18	
TE	299.6 K	21	5.1118	2044.0555	0.19	
D1	0.10000000 sec	22	5.1072	2042.2161	0.40	
TD0	1	23	5.1029	2040.4967	0.4/	
		24	5.0988	2038.8572	0.17	
	CHANNEL fl =======	25	5.0858	2033.6589	0.18	
NUC1	1H	26	5.0814	2031.8995	0.38	
P1	11.20 usec	27	5.0768	2030.0601	0.45	
PL1	6.00 dB	28	5.0728	2028.4606	0.15	
SF01	399.8724694 MHz	29	4.1987	1678.9342	0.56	
		30	4.1941	1677.0948	1.23	
F2 - Proc	essing parameters	31	4.1898	1675.3754	0.66	
SI	16384	32	4.1872	1674.3357	0.86	
SF	399.8700088 MHz	33	4.1829	1672.6163	1.13	
WDW	no	34	4.1781	1670.6969	0.69	
SSB	0	35	1.5842	633.4741	0.29	
LB	0.00 Hz	36	0.9354	374.0384	1.14	
GB	0	37	0.9285	371.2793	20.00	111
PC	1.00	38	0.9269	370.6395	9.67	M
		39	0.9214	368.4402	1.11	
		40	0.9105	364.0816	0.20	
		41	0.8963	358.4035	0.14	
		42	0.8783	351.2058	0.24	6 0
		43	0.0941	37.6278	0.53	0.0
		44	0.0869	34.7487	14.04	
		45	0.0854	34.1489	7.16	
		46	0.0790	31.5897	0.69	







32077-152 fr 5-7 nmr400b h-1

Current NAME	Data Parameters 32077-152														nn
EXPNO PROCNO	3 1				77		1 3			ы м ч		0			u v
F2 - Ac Date_ Time INSTRUM PROBHD PULPROG TD SOLVENT NS DS	cquisition Parameters 20110409 13.56 5 mm QNP 1H/1 5 cgdc 65536 CDC13 835 4						114.			77.5		64.3			ר ער ער
SWH FIDRES AQ RG DW DE TE D1 d11 TD0	26315.789 Hz 0.401547 Hz 1.2452340 sec 8192 19.000 usec 7.00 usec 300.1 K 0.10000000 sec 0.03000000 sec 40													0. 	
====== NUC1 P1 PL1 SF01	= CHANNEL II ======= 13C 4.00 usec 0.00 dB 100.5584512 MHz														
CPDPRG2 NUC2 PCPD2 PL2 PL12 SFO2	= CHANNEL f2 ======= waltz16 1H 100.00 usec 120.00 dB 24.50 dB 399.8719994 MHz														
F2 - Pr SI SF WDW SSB LB GB PC	rocessing parameters 32768 100.5473690 MHz EM 0 1.00 Hz 0 1.40														
	Peak?(F1)1137.76552114.1298377.5478477.2301576.9122664.3233726.1589818.63039-5.0347	[ppm] 138 114 7797 7765 7733 6467 2630 1873 -506	?(F1) 51.9586 75.4511 .2273 .2834 .3194 .5386 .2086 .2276 .2258	[Hz] 3.40 4.21 5.50 5.70 5.70 5.03 15.00 1.47 7.12	Intensity		1								
lafuu's agusta shuara ny isad ing u Tyd	ny blanky y stala gana a hiyo yang bana ankan shara sana daga sana yang yang bana		γγαληλή _{α σ} ατιμήτηλη αυτίρη (αυτήρας)	Mar Turn, ay for tagen in the Space of the Ass				1414-141-1414-1414-1414-1414-1414-1414	19-eniority:Mishballonity:Sta		And Market Street St			410017-10-10-10-10-10-10-10-10-10-10-10-10-10-	topon for the second
210	200 190 1	80 170	160	150 1	40 130	120	110	100	90	80	70	60	 50	40	30

32077-152 allylOTBS fr 5-7 nmr400b c-13

26.16 18.63 -5.03

]	Peak	?(F1)	[ppm]	?(F1)	[Hz]
			1	9.7471	4874	1.8172	1.01
			2	9.7439	4873	3.2168	1.10
Current Da	ata Parameters		3	7.2704	3636	5.1452	0.62
NAME	32077-169		4	3.8802	1940	0.6045	0.35
EXPNO	8		5	3.8698	1935	5.4031	0.37
PROCNO	1		6	3.8597	1930	.3518	0.76
			7	3.8493	1925	5.1504	0.77
F2 - Acqui	isition Paramete	ers	8	3.8324	1916	5.6982	0.75
Date	20110929		9	3.8197	191().3466	0.75
Time	16.54		10	3.8119	190	06.4456	0.30
INSTRUM	spect		11	3.7993	190	0.1439	0.31
PROBHD !	5 mm ONP 1H/13		12	2.5563	12	78.4823	0.11
PULPROG	zq30		13	2.5528	12	76.7319	0.09
TD	65536		14	2.5456	12	73.1309	0.09
SOLVENT	CDC13		15	2.5424	12	71.5305	0.19
NS	32		16	2.5396	12	70.1302	0.11
DS	4		17	2.5320	120	56.3292	0.11
SWH	13020.833 H	Hz	18	2.5293	120	54.9788	0.20
FIDRES	0.198682 4	Hz	19	2.5262	120	53.42.84	0.10
AQ	2.5166323 \$	sec	20	2.5190	12	59.8275	0.09
RG	228.1		21	2.5156	125	58.1270	0.11
DW	38.400 ι	usec	22	1.1059	553	3.0938	3.19
DE	6.50 ι	usec	23	1.0920	546	5.1420	3.09
TE	300.0 H	K	2.4	1.0868	543	3.5413	0.06
D1	0.1000000 \$	sec	25	1.0096	504	1.9313	0.06
TDO	1		2.6	0.9266	463	3,4205	0.15
			27	0.9223	461	.2699	0.27
======= (CHANNEL fl =====	====	28	0.9039	452	2.0675	0.12
NUC1	1H		29	0.8991	449	9.6669	0.08
P1	12.00 ι	usec	30	0.8957	44	7.9664	0.12
PL1	-4.00 0	dB	31	0.8921	446	5.1660	0.76
SF01	500.1330885 N	MHz	32	0.8864	443	3.3152	20.00
			33	0.8826	441	L.4147	0.26
F2 - Proce	essing parameter	rs	34	0.8804	440	0.3145	0.57
SI	32768		35	0.7595	379	9.8487	0.07
SF	500.1300083 N	MHz	36	0.1202	60	1156	0.08
WDW	no		37	0.0674	33.	7088	0.20
SSB	0		38	0.0622	31	.1081	5.53
LB	0.00 4	Hz	39	0.0561	28	.0573	0.18
GB	0					-	
PC	1.00						







Intensity

32077-169 branched fr 16-26 nmr500c h-1

NAME EXPNO PROCNO Date_ Time INSTRUM PROBHD TD TD SOLVENT NS DS	32077-169 9 1 20110929 17.08 spect 5 mm QNP 1H/13 zgdc 131072 CDC13 992 4	
SWH FIDRES AQ PC	40322.582 Hz 0.307637 Hz 1.6253552 sec	
DW DE TE D1	12.400 usec 6.50 usec 300.0 K 0.10000000 sec	
D11 TD0	0.03000000 sec 40	
======= NUC1 P1	CHANNEL f1 ======= 13C 2.50 usec	
PL1 SFO1	0.00 dB 125.7703648 MHz	
CPDPRG2 NUC2	CHANNEL f2 ======= waltz16 1H	
PCPD2 PL2 PL12 SE02	80.00 usec 120.00 dB 11.50 dB 500 1325007 MHz	
SI SF WDW	65536 125.7577615 MHz EM	
SSB LB GB PC	0 1.00 Hz 0	
	T • 10	

77.48 77.22 76.97

- 63.67

200

.

......

160

150 140 130

120

110

...... 100

.......

60

32077-169 branched fr 16-26 nmr500c c-13



Current I NAME EXPNO PROCNO F2 - Acqu Date_ Time INSTRUM	Data Parameters 32077-165 7 1 uisition Parameters 20110917 8.13 spect	Peak 1 2 3 4 5 6 7 8 9 10 11 12	?(F1) 7.3337 7.3270 7.3225 7.3063 7.2705 7.2647 7.2590 7.2580 7.2543 7.2543 7.2523 7.2472	[ppm] 3667.8 3664.4 3662.7 3654.0 3636.1 3633.2 3630.4 3629.9 3628. 3627. 3624.	?(F1) 034 526 521 020 999 952 945 437 436 0931 0929 5422	[Hz] 2.79 1.06 1.12 1.03 0.16 0.54 0.14 0.15 0.16 0.16 0.16 0.22	Intensity			(CH ₂ NH 0 4	CH₂Ph ∖Si │	
PROBHD PULPROG TD SOLVENT NS DS SWH FIDRES AQ RG DW DE TE D1 TD0 ======= NUC1 P1 PL1 SF01 E2 = Proc	5 mm QNP 1H/13 zg30 65536 CDC13 32 4 13020.833 Hz 0.198682 Hz 2.5166323 sec 114 38.400 usec 6.50 usec 300.0 K 0.10000000 sec 1 CHANNEL f1 ====== 1H 12.00 usec -4.00 dB 500.1330885 MHz	Peak 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36	?(F1) [3.8312 3.8203 3.8045 3.7936 3.7916 3.7651 3.5602 3.5490 3.5405 3.5297 3.5171 3.5101 3.4973 2.6980 2.6845 2.6749 2.6614 2.5212 2.5090 2.4980 2.4859 1.9089 1.9089 1.8834	ppm] 1916.09 1910.64 1902.74 1897.29 1896.29 1883.03 1780.56 1774.96 1770.71 1765.30 1759.00 1755.50 1749.10 1349.35 1342.59 1331.04 1260.92 1254.82 1243.27 954.698 948.246 941.944	?(F1) [1 81 67 46 32 29 95 52 99 55 29 90 73 66 89 90 77 8 60 60 778 60 60 778 60 8 92 92 95 55 99	Hz] 0.10 0.06 0.04 1.55 1.38 0.06 0.17 0.21 0.66 1.25 0.73 0.18 0.20 0.38 0.39 0.46 0.49 0.41 0.51 0.37 0.41 0.05 0.12 0.20	Intensity						
SI SF WDW SSB LB GB PC	32768 500.1300082 MHz no 0 0.00 Hz 0 1.00	37 38 39 40 41 42 43 44	1.8704 1.8580 1.8452 1.6845 0.9174 0.9039 0.8897 0.0452	935.443 929.241 922.835 842.465 458.815 452.067 444.965 22.6055	22 6 99 90 93 55 57 9	0.20 0.12 0.05 0.07 3.04 2.86 20.00 13.29	3.8	8 3.6	3.4 3.2	3.0	2.8 2.6	2.4 2.2	2.0
								0.95	0.90	0.85	ppm		,
					· · · ·					2.142	2.084	1.105	
9.5	9.0 8.5	8.0	7.5	7.0	6.5	6.0	5.5	5.0	4.5	4.0	3.5	3.0 2.5	2.0

32077-165 (R) chromatograpy fr 4-5 nmr500c h-1



NAME 32077-165 EXPNO 8 PROCNO 1 Date_ 20110917					(R) fr nmr	4-5 500c c-13
Time 8.22 INSTRUM spect PROBHD 5 mm QNP 1H/13 PULPROG zgdc TD 131072 SOLVENT CDC13 NS 3660 DS 4 SWH 40322 582 Hz	140.87	128.52 128.52 126.98	77.48	67.52	54.47	36.14
FIDRES 0.307637 Hz AQ 1.6253552 sec RG 8192 DW 12.400 usec DE 6.50 usec TE 300.0 K D1 0.10000000 sec D1 0.03000000 sec TD0 40				(₂Ph
====== CHANNEL f1 ======= NUC1 13C P1 2.50 usec PL1 0.00 dB SF01 125.7703648 MHz					4	
====== CHANNEL f2 ====== CPDPRG2 waltz16 NUC2 1H PCPD2 80.00 usec PL2 120.00 dB PL12 11.50 dB SF02 500.1325007 MHz SI 65536 SF 125.7577639 MHz WDW EM SSB 0 LB 1.00 Hz GB 0 PC 1.40						
 		······································	·····		······	
						. I

210

200

32077-165 (R) chromatography fr 4-5 nmr500c c-13





Current I NAME EXPNO PROCNO F2 - Acqu Date_ Time INSTRUM PROBHD PULPROG TD SOLVENT NS DS SOLVENT NS DS SWH FIDRES AQ RG DW DE TE	Data Parameters 32077-169 5 1 1 1 1 1 1 1 1 1 1 1 1 1	Hz Hz sec usec K	Peak 1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22	?(F1) 9.7951 9.7909 9.7865 7.2706 3.6983 3.6834 3.6691 3.6542 3.6393 2.5238 2.5238 2.5193 2.5059 2.5017 2.4883 2.4840 2.4667 2.4487 1.8940 1.8790 1.8765 1.8635 1.8616	[ppm] ?(F1) 3921.2724 3919.5910 3917.8296 2910.6393 1480.5405 1474.5755 1468.8508 1462.8859 1456.9210 1010.3529 1008.5514 1003.1870 1001.5056 996.1411 994.4197 987.4940 980.2881 758.2250 752.2201 751.2193 746.0150 745.2543	[Hz] 0.44 0.92 0.41 0.21 0.10 0.19 0.83 1.46 0.81 0.40 0.38 0.85 0.83 0.41 0.50 0.20 0.10 0.23 0.47 0.48 0.29 0.80
TD0 ======	1 CHANNEL f1 ====		24 25 26	1.8466 1.8438 1.8288	739.2494 738.1285 732.1235	0.44 0.41 0.20
NUC1	1H 12 75	11000	27	0.9079	363.4596	0.10
PL1	-2.00	dB	28 29	0.9049	362.2586 359 4563	0.17
PL1W SFO1	13.05791473 400.3324722	W MHz	30 31	0.8889	355.8533 352.9710	15.00 1.16
F2 - Proc SI SF WDW SSB LB GB PC	cessing paramete 16384 400.3300038 no 0 0.00 0.00 0 1.00	ers MHz Hz	32 33 34 35 36	0.0624 0.0572 0.0524 0.0447 0.0371	24.9806 22.8989 20.9773 17.8948 14.8522	1.71 0.30 0.47 10.14 0.31

1.00







Intensity

32077-169 linear fr 29-38 nmr400b h-1

NAME EXPNO PROCNO Date_ Time INSTRUM PROBHD TD SOLVENT NS	32077-169 14 1 20111125 10.56 spect 5 mm PABBO BB- 2gdc 5 5536 CDC13 1308
SWH	26315.789 Hz
FIDRES	0.401547 Hz
AQ	1.2452340 sec
RG	8192
DW	19.000 usec
DE	6.50 usec
TE	673.2 K
D1	0.10000000 sec
D11	0.03000000 sec
TD0	40
=======	CHANNEL f1 =======
NUC1	13C
P1	3.50 usec
PL1	0.00 dB
PL1W	31.90095711 W
SF01	100.6741319 MHz
CPDPRG2 NUC2 PCPD2 PL2 PL12 PL2W PL12W SFO2 SI SF WDW SSB LB GB PC	CHANNEL f2 ====== waltz16 1H 80.00 usec 120.00 dB 17.00 dB 0.00000000 W 0.16438942 W 400.3320017 MHz 32768 100.6630386 MHz EM 0 1.00 Hz 0 1.40

Annotation

210	200	190	180	170	160	150	140	130	120	110	100	90	80	70	60	50	40



-40.99

32077-169 linear nmr400b c-13



 	*****	 	L.,	

....